

Avogadro's Law:

$$\frac{V_1}{n_1} = \frac{V_2}{n_2}$$

When temperature and pressure are constant, this equation is used to compare different conditions on the same substance

$$V \propto n$$

Volume is directly proportional to the amount of gas (in moles)

$$\frac{V}{n} = k$$

Volume divided by the amount of gas equals the k constant
(k = proportionality constant)

V = Volume
n = Amount of Gas (moles)

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