Chemistry 111 Practice Problems 11

Material Covered

Chapter 9 (sections 1 and 2) and Chapter 23 (section 2 - only classes of organic compounds)

Suggested Problems

Chapter 9: 7, 9, 11, 13, 17, and 19 Chapter 23: 7

 Circle and name the functional group(s) present in each of the following molecules. Notice that lone pairs are omitted in the structures given below.

0 (ж) ОН-ОН-ОН 0 (d) НО-СОН-NH, (d) В-ОН-ОНОН

- Determine the electron geometry, molecular geometry, and idealized bond angles for each of the following molecules. In which cases do you expect deviations from the idealized bond angle?
 - disk tocks
 - (b) CSo
 - 60 NH,Br
- 3. I odine forms three molecules with chlorine: I/O, I/O_b, and I/O_b. Draw the Lewis structure for each molecule.
 - (a) Determine the molecular geometry of ICI.
 - (b) Determine the electron geometry and molecular geometry of ICI, and ICI,
- 4. Which would have the greater O-N-O bond angle, NO₅ or NO₅ ? Defend your answer.
- 5. Consider the following molecules: H₅S, Pf₈, CS₉, IBr, and CBr₈. Which molecules in this list are not polar? Justify your reasoning.
- Determine whether each of the following molecules is polar or nonpolar. Recall that the shape of a molecule is not necessarily given by the Lewis structure. (Notice that lone pairs are omitted in the following structures.)

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Teach Students How to Learn Saundra Yancy McGuire, 2023-07-03 Co published with NISOD Miriam a freshman Calculus student at Louisiana State University made 37 5% on her first exam but 83% and 93% on the next two Matt a first year General Chemistry student at the University of Utah scored 65% and 55% on his first two exams and 95% on his third These are representative of thousands of students who decisively improved their grades by acting on the advice described in this book What is preventing your students from performing according to expectations Saundra McGuire offers a simple but profound answer If you teach students how to learn and give them simple straightforward strategies to use they can significantly increase their learning and performance For over a decade Saundra McGuire has been acclaimed for her presentations and workshops on metacognition and student learning because the tools and strategies she shares have enabled faculty to facilitate dramatic improvements in student learning and success This book encapsulates the model and ideas she has developed in the past fifteen years ideas that are being adopted by an increasing number of faculty with considerable effect The methods she proposes do not require restructuring courses or an inordinate amount of time to teach They can often be accomplished in a single session transforming students from memorizers and regurgitators to students who begin to think critically and take responsibility for their own learning Saundra McGuire takes the reader sequentially through the ideas and strategies that students need to understand and implement First she demonstrates how introducing students to metacognition and Bloom's Taxonomy reveals to them the importance of understanding how they learn and provides the lens through which they can view learning activities and measure their intellectual growth Next she presents a specific study system that can quickly empower students to maximize their learning Then she addresses the importance of dealing with emotion attitudes and motivation by suggesting ways to change students mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Saundra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory Affective Dimensions in Chemistry Education Murat Kahveci, Mary Kay Orgill, 2015-01-08 This is a unique resource for those wishing to address the affective domain as they research and solve problems in chemistry education Contributions by world leading experts cover both fundamental considerations and practical case studies This work fills a gap in the literature of chemistry education which so far has focussed mainly on the cognitive

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