

Chemical Reactions and Balancing Chemical Equations Web Quest

Name: _____ Date: _____ Class: _____

Chemical Equations

http://web.archive.org/web/20131203110147/http://www.files.chem.vt.edu/RVGS/ACT/notes/Typos_of_Equations.html

Click "Directions"

- What three things does a balanced equation show you?
 - The _____ which enter into a reaction.
 - The _____ which are formed by the reaction.
 - The amounts of each substance _____ and each _____ produced.
- What two things must we remember when balancing equations?
 - Every chemical compound has a _____ which cannot be _____.
 - A chemical _____ must account for every _____ that is used, which is an application of the Law of _____ of _____.
- What does the \longrightarrow mean? _____
- What does the \longleftrightarrow mean? _____

Classic ChemBalancer –

<http://funbasedlearning.com/chemistry/chemBalancer/default.htm>

(1) Click the button for "Directions" and **read carefully**. Click the "OK" button and return to the game screen.

(2) Click "Start Game" button to give it a try!

(3) Start by adding a "1" in each box and compare the number of atoms of each element you have on each side of the equation.

(4) Change coefficients to balance each equation and click the "Balanced" button to check it. If it is wrong, correct it!

(5) Use the information in the pop-up windows to answer each question and then write the balanced equation before clicking the "OK" button.

#1

What does "ferrum" mean? _____
What color is sulfur? _____

#2

What is HCl? _____
Where is it found in the body? _____

#3

What are pyrotechnics? _____

#4

What was the Hindenberg? _____
What gas was used in it? _____
What gas is used today? _____

Chemical Reactions Web Quest Answer Key

Bowker Editorial Staff



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Elements of Chemical Reaction Engineering EduGorilla Prep Experts,2024-06-09 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels

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The Software Encyclopedia 2000 Bowker Editorial Staff,2000-05

Chemical Principles Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry s frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding

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