

*UNIT 2: In-class and HW problems***Lewis structures: Step-by-step method**

- a) Determine the total number of valence electrons in your molecule:
(i) Consider **negative** and **positive** ions; (ii) This number **does not** change!!
- b) Determine how atoms are connected:
(i) Least electronegative atom is usually the center, (ii) Each bond contains 2 electrons
(iii) Only use **single** bonds initially
- c) Complete the octet of outer atoms. Place electrons as pairs (use dots)
- d) Complete octet of central atom:
(i) Too few electrons - use multiple bonds (ii) Too many electrons - place on central atom
- e) Check resonance and/or formal charge:
(i) Non-equivalent bonds to equivalent atoms? (ii) Formal charges minimized?

In-class problems (most will be done during class)

1. Which of the following ionic compounds would have the largest lattice energies?
(a) LiCl or MgCl₂ (b) KF or KBr (c) MgO or SrO
2. (a) Which bond is the most polar? H-F, H-Cl, H-O
(b) Which bond is most ionic? Na-F, K-F, K-Br
3. (a) Give Lewis structures for the following molecules and ions. Show resonance structures if necessary. (b) Describe electron group geometry and molecular shapes using VSEPR rules.

HF	H ₂ CO	OF ₂	HCO ₂ ⁻	PCl ₅
H ₂ S	CO	NO ₃ ⁻	SF ₆	XeF ₄
NH ₃	CO ₂	NO	I ₃ ⁻	O ₃
CF ₄	NH ₄ ⁺	CH ₃	SOCl ₂	ClF ₃
4. Several possible Lewis structures can be drawn for the thiocyanate ion, NCS⁻. Predict the best Lewis structure for the thiocyanate ion based upon formal charge considerations:

Chemistry 131 Lab Manual Answers

Minjie Lin



Chemistry 131 Lab Manual Answers:

This book delves into Chemistry 131 Lab Manual Answers. Chemistry 131 Lab Manual Answers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemistry 131 Lab Manual Answers, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Chemistry 131 Lab Manual Answers
- Chapter 2: Essential Elements of Chemistry 131 Lab Manual Answers
- Chapter 3: Chemistry 131 Lab Manual Answers in Everyday Life
- Chapter 4: Chemistry 131 Lab Manual Answers in Specific Contexts
- Chapter 5: Conclusion

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