

PRACTICE EXAM 2

CHEM 1411 (Chap. 1, 2, 3, 4, 5)

- d 1. Which of the following is a physical change?
 - (a) rusting a nail
 - (b) fruits ripen
 - (c) baking a cake
 - (d) ice melts
 - (e) milk turns sour
- e 2. Which of the following is a homogeneous mixture?
 - (a) water and sand
 - (b) sand, alcohol, and sugar
 - (c) oil and vinegar
 - (d) milk of magnesium
 - (e) air
- e 3. Which of the following is a correct?
 - (a) sodium, S
 - (b) mercury, Mg
 - (c) tin, N, Sn
 - (d) potassium, P
 - (e) gold, Au
- b 4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?
 - (a) 41
 - (b) 314
 - (c) 378
 - (d) 457
 - (e) -32
- a 5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?

$d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$

 - (a) 1.59
 - (b) 1.74
 - (c) 2.36
 - (d) 2.89
 - (e) 3.50
- e 6. How many significant figures are there in the measurement of 1.0356×10^4 ?
 - (a) 1
 - (b) 2
 - (c) 3
 - (d) 4
 - (e) 5
- c 7. How many significant figures are appropriate to show in the result after carrying out the operation below?

$(13.7 + 0.027) + 8.221 = ?$ \times, \div least s.f. $+, -$ least decimal

13.727

 - (a) 1
 - (b) 2
 - (c) 3
 - (d) 4
 - (e) 5
- d 8. How many protons, electrons, and neutrons are there in $^{130}\text{Ba}^{2+}$?
 - (a) 56p, 56e, 130n
 - (b) 56p, 54e, 130n
 - (c) 56p, 58e, 74n
 - (d) 56p, 54e, 74n
 - (e) 54p, 56e, 74n
- c 9. Which of the following pairs would have similar chemical and physical properties?
 - (a) Na, Mg
 - (b) H, Li
 - (c) He, Ne
 - (d) Cr, Ca
 - (e) Cl, O
- c 10. Which of the following is the molecular formula?
 - (a) NaCl
 - (b) C_2F_4
 - (c) HCl
 - (d) AlBr_3
 - (e) NaCN
- e 11. Which of the following is the empirical formula?
 - (a) C_2H_4
 - (b) CH_3COOH
 - (c) P_2O_{10}
 - (d) NH_4NO_3
 - (e) Ag_2CO_3
- e 12. Which of the following is a correct match?
 - (a) BaCl_2 , barium dichloride
 - (b) CS_2 , carbon sulfide
 - (c) CuF_2 , copper (I) fluoride
 - (d) HNO_2 , nitrous acid
 - (e) $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, magnesium sulfate heptahydrate
- b 13. Which of the following pairs is a correct match?
 - (a) $\text{Ca}_3(\text{PO}_4)_2$, calcium (II) phosphate
 - (c) Mg_2O_3 , magnesium oxide
 - (b) $(\text{NH}_4)_2\text{SO}_4$, ammonium sulfate
 - (e) H_2CO_3 , carbon acid
 - (d) $\text{Cu}(\text{NO}_3)_2$, copper dinitrate

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Joseph William Mellor



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Synchrotron Radiation C. Kunz, 2013-11-11 Synchrotron radiation as a spectroscopic research tool has undergone a most interesting and astonishing historical development and has now come to the stage of an exciting boom The machines which produce synchrotron radiation were built and developed exclusively for other purposes in the past namely high energy physics At the same time however they involuntarily became better and better light sources for the spectral range from the visible to the hard x ray region Now we are at the point that the first few storage rings have gone into operation as machines dedicated to synchrotron radiation and several more are in the stage of construction and planning All this was brought about by the successful research performed during the past fifteen years in which several groups all over the world have participated at different accelerator centers mostly symbiotic with high energy physics As it happens with a young and rapidly developing field the number of reviews and monographs is still minute The objective of this book is to fill an apparent gap and to provide a sound basis for those who are interested in synchrotron radiation and its applications

Evaluated Kinetic Data for High Temperature Reactions D. L. Baulch, 1972 Publications of the National Bureau of Standards United States. National Bureau of Standards, 1976

NBS Special Publication, 1968 **Publications** United States. National Bureau of Standards, 1977 Evaluated Kinetic Data for High Temperature Reactions: Homogeneous gas phase reactions of the H₂-N₂-O₂ system, 1972 **EPA**

Publications Bibliography Quarterly Abstract Bulletin United States. Environmental Protection Agency, 1994 A Comprehensive Treatise on Inorganic and Theoretical Chemistry Joseph William Mellor, 1961 *Supplement to Mellor's Comprehensive Treatise on Inorganic and Theoretical Chemistry* Joseph William Mellor, 1961 *Reports on the Progress of Applied Chemistry* Society of Chemical Industry (Great Britain), 1962 *Supplement to Mellor's Comprehensive Treatise on Inorganic and Theoretical Chemistry: suppl. 3. K, Rb, Cs, Fr* Joseph William Mellor, 1961 *Electrogenerative Chemical Processes* Sergei Yurchak, 1969 Publications of the National Bureau of Standards ... Catalog United States. National Bureau of Standards, 1966 **EPA Publications Bibliography**, 1994 **Kiln-Drying of Lumber** R.B. Keey, T.A.G.

Langrish, J.C.F. Walker, 2012-12-06 At present no single book adequately covers a basic understanding of wood book satisfies the need for such a work It describes drying in practice This the fundamental basis of kiln drying technology to enable forest companies to improve their drying operations as high quality timbers become scarcer and of yesteryear can no longer be tolerated Adaptive the wasteful practices is no longer good enough Innovations change based on past experience of the material being dried and the processes require a sound understanding of drying Newer techniques such as the use of ultrahigh temperature sea soning and superheated steam under vacuum require an even greater depth of physical understanding for these methods to be used effectively and economically book provides a description of modern ideas about wood structure This moisture movement and stress development from which models of the drying process are developed to give the kiln operator important information about the course of drying under specified conditions and thus a means is compared with practice wherever for rational process improvement Theory possible **Biomarkers, Diagnostics and**

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