

AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%–9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$ (notice the d-block)
 - Nickel (with shorthand): $[\text{Ar}] 4s^2 3d^8$
- **Law of Definite Proportions:**
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

Chem Study Guide For Spring Benchmark

M Walker



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Water quality in the Great and Little Miami River Basins, Ohio and Indiana, 1999-2001 , *Energy Research Abstracts* ,1993 **Chemists' Guide to Effective Teaching** Norbert J. Pienta,Melanie M. Cooper,Thomas J. Greenbowe,2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry **Innovative Techniques for Large-group Instruction** National Science Teachers Association,2002 Size does matter When you re faced with a class of 50 150 or even 250 college students it s tough to head off boredom much less promote higher order thinking and inquiry skills But it s not impossible thanks to the professor tested techniques in this collection of 14 articles from the Journal of College Science Teaching The book starts by examining what research shows about the effectiveness of popular teaching styles Surprise Lectures don t stimulate active learning From there the authors offer proven alternatives that range from small scale innovations to completely revamped teaching methods Suggested strategies include using quizzes in place of midterms and finals student forums interactive lectures collaborative groups group facilitators and e mail and computer technology **The Publishers' Trade List Annual** ,1985 **Scientific and Technical Aerospace Reports** ,1985 *Resources in Education* ,1976 **Subject Guide to Books in Print** ,2001 *Stumbling Toward Sustainability* John C. Dernbach,2002 In 1992 at the United Nations Conference on Environment and Development in Rio de Janeiro the nations of the world agreed to implement an ambitious plan for ecologically sustainable human development This book is a comprehensive review of U S efforts to achieve such development since Rio The U S has unquestionably begun to take steps toward sustainable development Yet the nation is now far from being a sustainable society and in many respects is farther away than it was in 1992 Nevertheless legal and policy tools are available to put the U S on a direct path to sustainability This book brings together 42 distinguished experts from a variety of backgrounds and academic disciplines It is among the most thorough assessments ever conducted of U S law and policy concerning the environment *Proceedings of the Mid-year Meeting - Chemical Specialties Manufacturers Association* Chemical Specialties Manufacturers Association,1955 **Illinois Chemistry Teacher** ,2001 *Nuclear Science Abstracts* ,1973 NSA is a comprehensive collection of international nuclear science and technology literature for the period 1948 through 1976 pre dating the prestigious INIS database which began in 1970 NSA existed as a printed product Volumes 1 33 initially created by DOE s predecessor the U S Atomic Energy Commission AEC NSA includes citations to scientific and technical reports from the AEC the U S Energy Research and Development Administration and its contractors plus other agencies and international organizations

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