



Reinforcement

Chemical Changes

Read the passage and answer the following questions:



1. What is the chemical formula of hydrogen peroxide?

It is represented as the decomposition of hydrogen peroxide. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy in the form of a chemical bond.

2. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 . The chemical formula of hydrogen peroxide is H_2O_2 .

3. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 .

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in a variety of applications, including as a disinfectant, a bleaching agent, and a component of rocket propellants. It is also used in the chemical industry for the production of various chemicals.

4. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, the total mass of the reactants must be equal to the total mass of the products. Is this correct?

Yes, it is.



5. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 .

6. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 .

7. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 .

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in a variety of applications, including as a disinfectant, a bleaching agent, and a component of rocket propellants.

8. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 .

9. What is the chemical formula of hydrogen peroxide? It is represented as H_2O_2 .

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in a variety of applications, including as a disinfectant, a bleaching agent, and a component of rocket propellants.

10. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, the total mass of the reactants must be equal to the total mass of the products. Is this correct?

Chemical Changes Section 1 Reinforcement Chapter

Jay K. Hackett



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