

# Compound Names & Formulas

Determine whether the following compounds are bonded covalently or ionically and provide the proper formula or name for each of the compounds.

(This is a mixed review of the concepts in this lesson. Make sure that the appropriate rule for each type of compound is followed when naming.)

Chemical Name	Chemical Formula	Chemical Name	Chemical Formula
Aluminum hydroxide			$\text{Ag}_3\text{P}$
Zinc sulfide			$\text{BBr}_3$
Silicon tetrafluoride			$\text{C}_2\text{Br}_4$
Silver phosphate			$\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$
Nickel (III) sulfide			$\text{CaSO}_4$
Manganese (II) phosphate			$\text{Cr}(\text{CO}_3)_3$
Silver acetate			$\text{CuOH}$
Diboron tetrabromide			$\text{FePO}_4$

# Chemical Names And Formulas Packet Key

**G Psacharopoulos**



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