

Study Guide: Chemical Reaction

Name: _____

916

I. Physical and Chemical Change

Change is the act of altering a substance. An event in which there is a "Before" and "After" condition. Changes can be **physical** or **chemical**. Analyse the comparison chart and then, do the exercise.

Physical Change	Chemical Change
When a substance undergoes a physical change, its composition remains the same despite its molecules being rearranged in space.	When a substance undergoes a chemical change, its molecular composition is changed entirely. Thus, chemical changes involve the formation of new substances.
A Physical change affects only physical properties i.e. shape, size, etc.	Chemical change both physical and chemical properties of the substance including its composition.
A physical change involves very little to no absorption of energy.	During a chemical reaction, absorption and evolution of energy take place.
Generally, physical changes do not involve the production of energy.	Chemical changes usually involve the production of energy (which can be in the form of heat, light, sound, etc.)
Physical change is easily reversible i.e. original substance can be recovered.	Chemical changes are irreversible i.e. original substance cannot be recovered.

ACTIVITY

According to the chart above, mention all the examples you can think of for each type.

Physical Change	Chemical Change

Chemical Reactin Study Guide Answers

Julie R. Frentrup



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