

Identifying Reaction Types & Balancing Equations

<u>Step 1:</u> Name the type of chemical reaction	<u>Step 2:</u> Balance the following chemical equations:
1. Ex: Synthesis	$\text{___ N}_2 + \text{___ H}_2 \rightarrow \text{___ NH}_3$
2.	$\text{___ KClO}_3 \rightarrow \text{___ KCl} + \text{___ O}_2$
3.	$\text{___ NaCl} + \text{___ F}_2 \rightarrow \text{___ NaF} + \text{___ Cl}_2$
4.	$\text{___ H}_2 + \text{___ O}_2 \rightarrow \text{___ H}_2\text{O}$
5.	$\text{___ AgNO}_3 + \text{___ MgCl}_2 \rightarrow \text{___ AgCl} + \text{___ Mg(NO}_3)_2$
6.	$\text{___ AlBr}_3 + \text{___ K}_2\text{SO}_4 \rightarrow \text{___ KBr} + \text{___ Al}_2(\text{SO}_4)_3$
7.	$\text{___ CH}_4 + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
8.	$\text{___ C}_3\text{H}_8 + \text{___ O}_2 \rightarrow \text{___ CO}_2 + \text{___ H}_2\text{O}$
9.	$\text{___ FeCl}_3 + \text{___ NaOH} \rightarrow \text{___ Fe(OH)}_3 + \text{___ NaCl}$
10.	$\text{___ P} + \text{___ O}_2 \rightarrow \text{___ P}_2\text{O}_5$
11.	$\text{___ Na} + \text{___ H}_2\text{O} \rightarrow \text{___ NaOH} + \text{___ H}_2$
12.	$\text{___ Ag}_2\text{O} \rightarrow \text{___ Ag} + \text{___ O}_2$
13.	$\text{___ CO}_2 + \text{___ H}_2\text{O} \rightarrow \text{___ C}_6\text{H}_{12}\text{O}_6 + \text{___ O}_2$
14.	$\text{___ K} + \text{___ MgBr}_2 \rightarrow \text{___ KBr} + \text{___ Mg}$
15.	$\text{___ HNO}_3(\text{aq}) + \text{___ Ba(OH)}_2(\text{aq}) \rightarrow \text{___ Ba(NO}_3)_2(\text{aq}) + \text{___ H}_2\text{O (l)}$
16.	$\text{___ C}_5\text{H}_{12}(\text{g}) + \text{___ O}_2(\text{g}) \rightarrow \text{___ CO}_2(\text{g}) + \text{___ H}_2\text{O(g)}$
17.	$\text{___ Al(s)} + \text{___ Fe}_2\text{O}_3(\text{aq}) \rightarrow \text{___ Al}_2\text{O}_3(\text{aq}) + \text{___ Fe(s)}$
18.	$\text{___ Al(s)} + \text{___ O}_2(\text{g}) \rightarrow \text{___ Al}_2\text{O}_3(\text{s})$

Chemical Reactions Guided Study Work Answers

William K. Purves



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Study Guide and Solutions Manual Bruce A. Chase, Peter J. Russell, 2005-06 This student resource contains chapter outlines of text material solutions to all end of chapter problems key terms suggestions for analytical approaches problem solving strategies and a variety of additional questions for student practice Also featured are questions that relate to chapter specific animations and iActivities

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Exercise Physiology John Porcari, Cedric Bryant, Fabio Comana, 2015-02-25 Learn how to apply the science of exercise physiology to your exercise programs and to solve the problems you'll encounter every day in practice You'll explore the principles of movement on which exercise is based while you develop the confidence you need to create individualized exercise programs based on current lifestyles schedules and abilities and properly progress those fitness programs through the stages of the ACE IFT training model

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