D. neutron

1. What is called the building block of matter?

A. Atom B. compound C. molecules

Which has almost the same mass with the proton?

A. electron B. neutron C. nuclei D. orbital

,, Mg "

Answer 3-4 based from the atomic notation above.

3. How many protons does magnesium have?

A. 12 B. 24 C. +2 D. no answer

4. How many electron are there in Magnesium?

A. 12 B. 24 C. +2 D. no answer

5. A neutral atom has the same number of proton and electron. Which atomic notation below is correct showing equal number of electron and proton?

A. Na B. Na C. Na D. Na

If Magnesium has an atomic number of 12 and it has a charge of +2. Which subatomic particle is affected?

A. electron B. neutron C. proton D. none

7. Atomic mass is the total number of proton and neutron. Which do you think is the heaviest among the following elements?

A. Calcium B. Carbon C. Nitrogen D. Oxygen

8. Recalling the elements, which do you think has the most number of protons and least number of electrons among the elements below?

L Helium II. Magnesium III. Oxygen IV. Sodium V. Sulfur A. II. I B. III. I C. IV. I D. V. I

9. How is the atomic mass computed?

A. sum of electrons and protons

- B. sum of protons and neutrons
- C. sum of neutrons and electrons

D. sum of electrons, neutron and electrons

10. Which total mass is the smallest?

A. The mass of 2 neutrons

- B. The mass of 2 electrons
- C. The mass of 1 neutron plus mass of 1 proton
- D. The mass of 1 electron plus the mass of 1 proton
- 11. An atom has an atomic number of 48, and a mass number of 120

How many protons, electrons and neutrons does the atom have?

- A. The atom has 72 protons, 48 electrons, and 48 neutrons.
- B. The atom has 120 protons, 48 electrons, and 72 neutrons.
- C. The atom has 72 protons, 72 electrons, and 48 neutrons.
- D. The atom has 48, protons, 48 electrons, and 72 neutrons.
- 12. Which of the following is an element?

A. amalgam B. coal C. silica D. zirconium

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