

# BINARY IONIC COMPOUNDS WORKSHEET

Write the formulas for the following binary ionic compounds.

1. Aluminum Fluoride \_\_\_\_\_
2. Copper (I) Oxide \_\_\_\_\_
3. Ammonium Chloride \_\_\_\_\_
4. Iron (III) Fluoride \_\_\_\_\_
5. Copper (I) Sulfide \_\_\_\_\_
6. Titanium (III) Bromide \_\_\_\_\_
7. Mercury (I) Bromide \_\_\_\_\_
8. Tin (II) Selenide \_\_\_\_\_
9. Tin (IV) Phosphide \_\_\_\_\_
10. Gallium Arsenide \_\_\_\_\_
11. Gold (I) Sulfide \_\_\_\_\_
12. Rubidium Dichromate \_\_\_\_\_
13. Manganese (III) Sulfite \_\_\_\_\_
14. Chromium (III) Hydroxide \_\_\_\_\_
15. Tin (IV) Oxide \_\_\_\_\_

# Binary Ionic Homework Answer Keys

**IM Harris**



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