

SECTION - B

Note : There are 12 Short Answer Type Questions from Q. 3 to Q. 14 carrying two marks each. You have to attempt any eight questions. This section includes ten theory questions and two numerical problems. Write appropriate units for each calculated value.

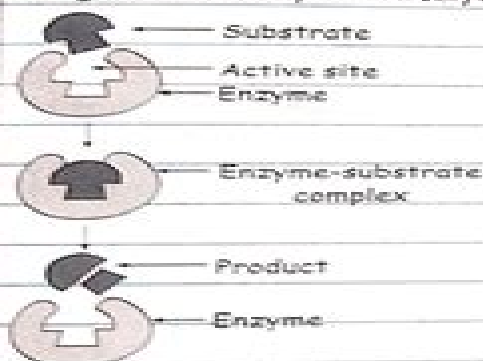
- Q. 3. (a) It oxidises lead sulphide (PbS) to lead sulphate (PbSO_4) changing oxidation state of S from -2 to $+6$.



- (b) Ozone oxidises nitrogen oxide (nitric oxide) to nitrogen dioxide.



- Q. 4. Diagram of enzyme catalysis :



(2 mark)

- Q. 5. If x_1 and x_2 are the mole fractions of solvent and solute respectively, then

$$x_1 + x_2 = 1$$

By Raoult's law,

$$P = x_1 \times P_0$$

where P^0 is the vapour pressure of a pure solvent and P is the vapour pressure of the solution at given temperature.

$$\therefore \frac{P}{P_0} = x_1$$

$$1 - \frac{P}{P_0} = 1 - x_1 \quad \therefore \frac{P_0 - P}{P_0} = x_2$$

$$\text{If } P_0 - P = \Delta P, \text{ then } \frac{\Delta P}{P_0} = x_2.$$

($\frac{1}{2}$ mark)

(1 mark)

Cybered Chemistry Answer Guide

BM King



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