Collision Theory Lab

Many people are familiar with the "plop, plop, fizz, fizz" that takes place when antacid tablets are dropped in a glass of water. As the tablet fizzes, it demonstrates a reaction that can be timed and studied. The fizz results when citric acid (H₂C₆H₂O₇) reacts with sodium bicarbonate (NaHCO₃), to give carbon dioxide, water, and a salt (sodium citrate).

All of the ingredients for this reaction are present in the antacid tablet, but nothing happens until you put it in water. The tablet, therefore, provides you with a nice chemical package that you can use to carry out timed kinetics experiments.

Temperature is one of many factors that affect the rate at which a chemical reaction proceeds. Collision theory states that as temperature increases, the particles in a reaction move faster, resulting in a higher frequency of collisions and increased kinetic energy in each collision. Particle size and concentration are two other factors that affect reaction rates.

In this lab, you will be given two antacid tablets and a challenge:

- Find a set of conditions in which one antacid tablet will react very quickly, OR
- Find a set of conditions in which the other antacid tablet will react very slowly.

The <u>REACTION TIME</u> for your challenge is defined as the time between when the antacid tablet COMES INTO CONTACT with water, and the when all FIZZING HAS STOPPED.

In order to carry out your task, you can use the following items.

- Timer, beaker, ice, stirring rod, hot plate, pestle and mortar, tongs, crucible, thermometer, graduated cylinder
- 2 antacid tablets—you MUST use these two and you will NOT be given more if you make a
 mistake
 - You will use one tablet as a <u>control</u>. The other tablet will be used for your <u>challenge</u>.

Control

Water temperature is approximately room	Summary of Conditions	Materials Used	Reaction Time (seconds)
approximately room	Whole tablet Water temperature is		
temperature ("C)	approximately room		

Describe your procedure below. You must write in full sentences.

Collision Theory Lab Report

John Orion Thomas,Bjørn Johan Landmark

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