

First Name _____ Last Name (Surname) _____ **Key**

1. (16 points) Using the reduction potentials provided, circle the appropriate choice in each case.

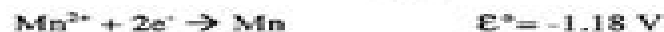
4 pts
each

Strongest Reducing Agent: VO^{2+} or Cr^{2+} (circle one)



See explanation
below

Strongest Oxidizing Agent: Mn^{2+} or Al^{3+} (circle one)



Strongest Reducing Agent: Cr or Zn (circle one)



Strongest Oxidizing Agent: Au^{3+} or Ag^+ (circle one)



2. (6 points) Determine the oxidation number in each case.



2 pts
each

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Table of Contents College Chemistry Questions And Answers

1. Understanding the eBook College Chemistry Questions And Answers
 - The Rise of Digital Reading College Chemistry Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying College Chemistry Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an College Chemistry Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from College Chemistry Questions And Answers
 - Personalized Recommendations
 - College Chemistry Questions And Answers User Reviews and Ratings
 - College Chemistry Questions And Answers and Bestseller Lists
5. Accessing College Chemistry Questions And Answers Free and Paid eBooks
 - College Chemistry Questions And Answers Public Domain eBooks
 - College Chemistry Questions And Answers eBook Subscription Services
 - College Chemistry Questions And Answers Budget-Friendly Options

6. Navigating College Chemistry Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - College Chemistry Questions And Answers Compatibility with Devices
 - College Chemistry Questions And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of College Chemistry Questions And Answers
 - Highlighting and Note-Taking College Chemistry Questions And Answers
 - Interactive Elements College Chemistry Questions And Answers
8. Staying Engaged with College Chemistry Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers College Chemistry Questions And Answers
9. Balancing eBooks and Physical Books College Chemistry Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection College Chemistry Questions And Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine College Chemistry Questions And Answers
 - Setting Reading Goals College Chemistry Questions And Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of College Chemistry Questions And Answers
 - Fact-Checking eBook Content of College Chemistry Questions And Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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