

Exercises (100)

Controlling Chemical Reactions

If the statement is true, write "true". If the statement does change the statement word or words to make the statement true.

1. Le Chatelier's Principle states that the equilibrium of a reaction shifts in the direction that counteracts the change.
2. True The concentration of a reactant in a closed system is the equilibrium of the reaction.
3. Le Chatelier's Principle states that the effect of a change in a chemical equilibrium is to counteract the change.
4. True Le Chatelier's Principle states that a change in the equilibrium of a reaction is to counteract the change.
5. True Le Chatelier's Principle states that a change in the equilibrium of a reaction is to counteract the change.

Put in the blank a complete word or words.

1. The reaction of a chemical system at equilibrium is dynamic.
2. In an equilibrium system, the change in the concentration of a reactant is the same as the change in the concentration of a product.
3. Increasing the concentration of a reactant in a chemical reaction shifts the equilibrium to the right.
4. The reaction of a chemical system at equilibrium is dynamic.
5. The reaction of a chemical system at equilibrium is dynamic.

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**Vermont Agricultural Experiment
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