### **Biological Molecules**

#### Organic and Inorganic compounds

- Organic compounds are based on \_\_\_\_\_ and are found in living things. There are a number of
  exceptions including hydrogen carbonate (HCO<sub>1</sub> ), carbon dioxide (CO<sub>2</sub> )and Carbon monoxide (CO).
- compounds are by default are all the molecules other than those in the category above.

#### Common organic molecules

Monosaccharide sugars. These are the \_\_\_\_\_\_\_ from which larger \_\_\_\_\_\_ molecules are constructed. Molecules like glucose and fructose are metabolically active molecules usually stored in an inactive, insoluble polysaccharide form.

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- <u>Glucose</u>: C<sub>0</sub>H<sub>12</sub>O<sub>6</sub> is a \_\_\_\_\_ sugar (six carbons) most commonly found in this ring structure.
- Glucose will be known to most students as a product of or the substrate molecule for
- Glucose is also found in a polymer form of starch, glycogen and cellulose.
- •All bonds are
- Glucose is a reducing sugar and will give positive (Brick red) precipitate in a test.
- ·Glucose is metabolically active compound
- Glucose is \_\_\_\_\_ and has osmotic effects when in solution.

Ribose: (5 carbon sugar).

(hydrogen bonding) when in solution.

A modified version of ribose, deoxyribose is perhaps best known for its role in Deoxyribonucleic acid or \_\_\_\_\_ where it forms part of the sugar phosphate backbone. The chemical properties of deoxyribose are very \_\_\_\_ from the properties of ribulose. Both Ribous and Glucose will attract water molecules.

Amino Acids: There are \_\_\_\_\_common amino acids found in the protein structures of living things. Amino acids are monomers which combine to form the larger \_\_\_\_\_, in turn polypeptides combine to form \_\_\_\_. Proteins molecules are the basis of \_\_\_\_ and many cellular and extra cellular components.



Amino
NH2 — C — COOH
carboxylic
H acid

This model shows the structure of the general amino acid. If you build one in a molecular kit you will appreciate better the 3D structure. Each of the common amino acids has the same structure as the one shown except that the R group is different. Amino acids are soluble.

This is an alternative way to draw the general amino acid structure. This diagram illustrates the 'amino' group which is -NH<sub>2</sub>. There is also the acidic group -COOH which ionizes in solution to form an -COO and H'groups. This acid group is known as a carboxylic acid group.

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