

Cp physics - Spring Final Review (second semester topics)**Multiple Choice**

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following is a direct cause of a substance's temperature increase?
- Energy is removed from the particles of the substance.
 - Kinetic energy is added to the particles of the substance.
 - The number of atoms and molecules in a substance changes.
 - The volume of the substance decreases.
- _____ 2. What happens to the internal energy of an ideal gas when it is heated from 0°C to 4°C ?
- It increases.
 - It decreases.
 - It remains constant.
 - It is impossible to determine.
- _____ 3. Which of the following is proportional to the kinetic energy of atoms and molecules?
- elastic energy
 - temperature
 - potential energy
 - thermal equilibrium
- _____ 4. As the temperature of a substance increases, its volume tends to increase due to
- thermal equilibrium.
 - thermal expansion.
 - thermal contraction.
 - thermal contraction.
- _____ 5. What is the temperature of a system in thermal equilibrium with another system made up of water and steam at 1 atm of pressure?
- 0°F
 - 273 K
 - 0 K
 - 100°C
- _____ 6. Energy transferred as heat occurs between two bodies in thermal contact when they differ in which of the following properties?
- mass
 - specific heat
 - density
 - temperature
- _____ 7. Which of the following terms describes a transfer of energy?
- heat
 - internal energy
 - temperature
 - kinetic energy
- _____ 8. The use of fiberglass insulation in the outer walls of a building is intended to minimize heat transfer through what process?
- conduction
 - radiation
 - convection
 - vaporization
- _____ 9. How is energy transferred as heat always directed?
- from an object at low temperature to an object at high temperature
 - from an object at high temperature to an object at low temperature
 - from an object at low kinetic energy to an object at high kinetic energy
 - from an object with higher mass to an object of lower mass
- _____ 10. What three properties of a substance affect the amount of energy transferred as heat to or from the substance?
- volume, temperature change, specific heat capacity
 - density, temperature change, specific heat capacity
 - mass, temperature change, specific heat capacity
 - mass, temperature change, latent heat

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Essential Advanced General Mathematics Third Edition Enhanced TIN/CP Version Michael Evans, Kay Lipson, Douglas Wallace, Sue Avery, 2011-04 The Essential VCE Mathematics series has a reputation for mathematical excellence with an approach developed over many years by a highly regarded author team of practising teachers and mathematicians This approach encourages understanding through a wealth of examples and exercises with an emphasis on VCE examination style questions New in the enhanced versions TI Nspire OS3 and Casio ClassPad calculator explanations examples and problems are integrated into the text Page numbers in the printed text reflect the previous TI nspire and Casio ClassPad version allowing for continuity and compatibility Digital versions of the student text are available in Interactive HTML and PDF formats through Cambridge GO

Chemistry for Degree Students B.Sc. Semester - II (As per CBCS) Madan R.L., 2017 This textbook has been designed to meet the needs of B Sc Second Semester students of Chemistry as per the UGC Choice Based Credit System CBCS With its traditional approach to the subject this textbook lucidly explains principles of chemistry Important topics such as chemical energetics chemical ionic equilibrium aromatic hydrocarbons alkyl aryl halides alcohols phenols ethers aldehydes and ketones are aptly discussed to give an overview of physical and organic chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures

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Boston University Bulletin Boston University, 1926

Catalogue for the Year ... Boston University. College of Liberal Arts, 1920

Student's Guide to Undergraduate Physics Major Departments American Institute of Physics, 1969

Johns Hopkins University Circulars Johns Hopkins University, 1945

Resources in Education ,1976

University of Cincinnati Bulletin ,1925

American Bee Journal ,1915 Includes summarized reports of many bee keeper associations

1998 Guide to the Evaluation of Educational Experiences in the Armed Forces American Council on Education, 1998 This guide is the standard reference work for recognizing learning acquired by military personnel for conversion to academic credit in degree work at colleges and universities This volume contains recommendations for formal courses offered by the Air Force the Coast Guard the Marine Corps and the Department of Defense in 1990 and later years Twenty sections provide the following information 1 how to find and use course exhibits 2 sample Coast Guard Rating exhibit 3 how to find and use Coast Guard occupation exhibits 4 sample Coast Guard rating exhibit 5 how to find and use Coast Guard warrant officer exhibits 6 sample Coast Guard warrant officer exhibit 7 how to find and use Marine Corps MOS military occupational specialties exhibits 8 sample Marine Corps enlisted MOS exhibit 9 questions and answers 10 awarding credit for extrainstitutional learning 11 transfer and award of credit 12 elements of a model policy on awarding credit for extrainstitutional learning 13 Air Force

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