

10.1 Periodic Trends



10.1 PERIODIC TRENDS FROM CONCEPTS

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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| _____ | 97 |
| _____ | 98 |
| _____ | 99 |
| _____ | 100 |

| | |
|------------------------|-------|
| Atomic Mass | _____ |
| Atomic Number | _____ |
| Electron Configuration | _____ |
| Chemical Name | _____ |
| Chemical Symbol | _____ |

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 1. It increases in the periodic table.
- _____ 2. It decreases in the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are shiny and conduct electricity.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, changing the underlined word as shown in *smaller* in *smaller* in *smaller*.

- _____ 1. *There are more representative elements of elements.*
- _____ 2. *There are more elements of the elements in the periodic table.*
- _____ 3. *Group 1 elements are highly reactive elements in the periodic table.*
- _____ 4. *Group 17 elements are the elements in the periodic table.*
- _____ 5. *Group 1 elements are highly reactive elements in the periodic table.*
- _____ 6. *Group 17 elements are highly reactive elements in the periodic table.*
- _____ 7. *Group 1 elements are highly reactive elements in the periodic table.*
- _____ 8. *Group 17 elements are highly reactive elements in the periodic table.*
- _____ 9. *Group 1 elements are highly reactive elements in the periodic table.*
- _____ 10. *Group 17 elements are highly reactive elements in the periodic table.*

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Chapter 6

Cary I. Sneider



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Chemical Pedagogy Keith S Taber, 2024-12-20 How should chemistry be taught in schools colleges and universities Chemical Pedagogy discusses teaching approaches and techniques the reasoning behind them and the evidence for their effectiveness The book surveys a wide range of different pedagogic strategies and tactics that have been recommended to better engage learners and provide more effective chemistry teaching These accounts are supported by an initial introduction to some key ideas and debates about pedagogy the science of teaching Chemical Pedagogy discusses how teaching innovations can be tested to inform research based practice Through this book the author explores the challenges of carrying out valid experimental studies in education and the impediments to generalising study results to diverse teaching and learning contexts As a result the author highlights both the need to read published studies critically and the value of teachers and lecturers testing out recommended innovations in their own classrooms Chemical Pedagogy introduces core principles from research into human cognition and learning to provide a theoretical perspective on how to best teach for engagement and understanding An examination of some of the more contentious debates about pedagogy leads to the advice to seek optimally guided instruction which balances the challenge offered to learners with the level of support provided This

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