

2021 WASSCE CHEMISTRY THEORY

1.

(a) Distinguish between molecular formula and structural formula.

(b) List three factors that determine the ionization energy of an atom.

(c) State two conditions necessary for the establishment of chemical equilibrium.

(d) Consider the following table.

<i>Element</i>	A	B	C
<i>Ionization energy kJ/mol</i>	619	518	594

(i) State which of the elements is the strongest reducing agent.

(ii) Give the reason for the answer stated in 1(d)(i).

(e) State Graham's law of diffusion.

(f) Consider the following salts:

$\text{Mg}(\text{NO}_3)_2$, CaCO_3 , Na_2SO_4 .

State which of the salts is/are:

(i) readily soluble in water

(ii) insoluble in water

(g) Classify each of the following products as addition polymer or condensation polymer:

(i) Protein

(ii) Perspex

(iii) Nylon

(h) Define atomic radius

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