Chemistry - Unit 11 Practice Te	est 1 Name:	
Must show all work and labels for credit		
B 1. A solution in which [H+] = 1.0 × 10	12 Marie 2	
a) acidic (b) bs	sic c) neutral	
A 2. A solution in which [OH-] = 1.0 × 10	0° M is ?	
a) peidic (OH-) = 1.0 × 10		
3. When acids and bases react, they alw	rays form and	
water/gases water/salt	b) hydronium/gases d) hydronium/salt	
B 4. Strong acids/bases will have K 1		
a) < (6/>	e) =	
B s. HCl +(H ₂ O) + H ₂ O* + Cl shows	the model for saids	and haces
a) Arrhenius	(b))Bronsted-Lowry	
c) Gay-Lussac	d) Dalton	
6. Acids produce hydrogen	and bases produce hydroxide	in aqueous solutions
a) atoms	b) molecules	L 1000 - 10 to 100 000 100 000 100 000 100 100 100 10
c) isotopes	(d) ons	
A 7. pOH literally stands for "	-	
(a) potential hydroxide	b) probable hydroxide	
c) potential hydrogen	 d) probable hydrogen 	
B sacids/bases do not comple	tely dissociate: less free ions, mon	e molecules
a) Strong b) W	eak c) Neutral	to analytical to the total to all.
9. Match the names of these common acids a	nd bases with the correct formula	<u> </u>
E_ Hydrochloric Acid	A. NH ₃	12-
F Phosphoric Acid	8. CH,00H or H	C ₂ H ₂ O ₂
I Hydroxide	C. HF	
H Nitric Acid	D. H ₂ CO ₃	
Sulfuric Acid	E. HCI	
Carbonic Acid	F. H ₂ PO ₄	
B Acetic Acid	G. H,SO,	
A Ammonia	H HNO3	
C_Hydrofluoric Acid	I OH	

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