

Unit 3 Review - Atomic Structure & Periodic Table

Part 1: Answer the following questions.

1. a. Which scientist created the first modern atomic theory? John Dalton
 b. What was his theory?
 1. All matter is made of extremely small particles called atoms
 2. All atoms of a given element are identical and atoms of a specific element are different from any other element
 3. Atoms cannot be created, divided into smaller particles or destroyed
 4. Different atoms combine in simple whole number ratios to form compounds
 5. In a chemical reaction, atoms are separated, combined or rearranged.
- c. What was his model of the atom called? Billiard Ball
- d. What was proven wrong and why?
 1. All atoms of a given element are identical – wrong b/c of the existence of isotopes
 2. Atoms cannot be divided into smaller particles – wrong b/c of the existence of electrons, protons and neutrons
2. a. Which scientist did the gold foil experiment? Rutherford
 b. What did he conclude from the gold foil experiment?
 1. Atoms are made up of mostly empty space
 2. Atoms have a small positive center call the nucleus
3. a. Who discovered the electron? J. J. Thomson
 b. What was the name of his experiment? Cathode Ray Tube Experiment
 c. What was his model of the atom called? Plum Pudding Model
4. What did Mendeleev do?
Arranged the periodic table in order of increasing atomic mass
5. a. Who arranged the **modern** periodic table? Moseley
 b. How is the modern periodic table arranged? Arranged in order of Atomic Number
6. a. What are the three subatomic particles? Give the symbols, charge, mass and location of each.

Particles	Symbol	Charge	Mass	Location
Proton	p^+	Positive	1 amu	Nucleus
Neutron	n^0	Neutral	1 amu	Nucleus
Electron	e^-	Negative	1/2000 amu	Electron Cloud

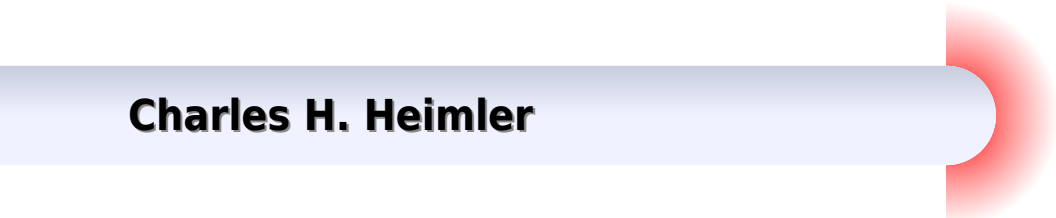
7. What is special about the proton? **The proton IDENTIFIES THE ELEMENT**
8. Fill in the blanks in the following:
 - a. Mass # = $p^+ + n^0$
 - b. Atomic # = # of protons
9. What is the difference between Average Atomic mass and mass #? (Average) Atomic Mass is found on the periodic table and is the weighted average of the isotopes. The Mass Number is the mass of one element and must be given or calculated by adding the protons and neutron of that element.
10. Which of the following are isotopes of the same element? How do you know? **A. and D. They have the same Atomic Number (# of Protons)**
 - a. $^{12}_5X$
 - b. $^{13}_5X$
 - c. $^{12}_6X$
 - d. $^{13}_5X$
11. Calculate the average atomic mass for Boron if it has two naturally occurring isotopes: boron -10 (abundance of 19.8%) and boron - 11 (abundance of 80.2%).

$$\left(10 \times \frac{19.8}{100} \right) + \left(11 \times \frac{80.2}{100} \right) = 10.802 \text{ amu}$$
12. Identify an element "X" by calculating the average atomic mass. X = 6 (abundance of 7.5%) and X= 7 (abundance 92.5%).

$$\left(6 \times \frac{7.5}{100} \right) + \left(7 \times \frac{92.5}{100} \right) = 6.925 \text{ amu} \quad \text{Which is Lithium}$$

Chemistry Unit 3 Periodic Table Test Answer

Charles H. Heimler



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wykorzystania numer nagrania symbol kategorii wyraźne hasła obrazki ułatwiające zapamiętywanie do wybranych haseł
fiszki niemiecki kurs i nauka języka niemieckiego dla - Jul 11 2022

web poziom a1 1040 fiszek a na nich 2100 słów i zwrotów dla rozpoczynających naukę języka niemieckiego grundstufe
dostępność w sprzedaży wysyłka w ciągu 24h 99 00 zł 79 00 zł słownictwo 2 niemiecki poziom a2 1040 fiszek a na nich 2300
słów i zwrotów dla początkujących grundstufe plus dostępność w sprzedaży wysyłka

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web oryginalne fiszki pozwalają szybko i skutecznie opanować ponad 2400 słów i zwrotów na poziomie zaawansowanym
wydanie zawiera 1040 trwałych kartoników 270 minut wyraźnych nagrań mp3 pudełko do nauki w formie quizu

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zakupów 100 bezpieczeństwa dla każdej transakcji kup teraz

niemiecki fiszki audio mp3 szybka nauka niemieckich słówek - Jun 10 2022

web dec 29 2020 fiszki audio mp3 język niemiecki kurs do szybkiej nauki niemieckiego słownictwa niemieckie fiszki mp3 to
sposób na efektywną i szybką naukę niemieckich słówek wejdź i pobierz fiszki na swój komputer i ucz się niemieckiego
słownictwa ze słuchu w każdej chwili

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web fiszki do nauki niemieckiego to małe karteczki które można kupić w sklepie wydrukować w domu albo zrobić
własnoręcznie po jednej stronie znajduje się słówko lub całe zdanie w języku polskim a druga to to samo ale już po niemiecku
osobiście uważamy że fiszki do nauki niemieckiego to jedna z fajniejszych metod

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web fiszki język niemiecki słownictwo 5 oryginalne fiszki pozwalają szybko i skutecznie opanować ponad 2400 słów i zwrotów
na poziomie zaawansowanym najnowsze wydanie zawiera 1040 trwałych kartoników 270 minut wyraźnych nagrań mp3
pudełko do nauki w formie quizu memobox program pc oraz praktyczne etui słowa kluczowe nauka

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seria cykl fiszki redakcja naukowa kinga perczyńska wydawca cztery głowy typ oprawy kartonowa produkt niedostępny dodaj do schowka zarezerwuj i odbierz w księgarni stacjonarnej pwn opis dane szczegółowe

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web słownictwo 5 cd opracowanie zbiorowe książka w sklepie empik com książka fiszki język niemiecki słownictwo 5 cd autorstwa opracowanie zbiorowe dostępna w sklepie empik com w cenie 17 35 zł przeczytaj recenzję fiszki język niemiecki słownictwo 5 cd zamów dostawę do dowolnego salonu i zapłać przy odbiorze

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web nov 9 2011 książka fiszki język niemiecki słownictwo 5 autorstwa opracowanie zbiorowe dostępna w sklepie empik com w cenie przeczytaj recenzję fiszki język niemiecki słownictwo 5 zamów dostawę do dowolnego salonu i zapłać przy odbiorze

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słownictwo 5 niemiecki pliki mp3 do pobrania fiszki pl - Dec 16 2022

web fiszki audio mp3 do pobrania j niemiecki słownictwo 5 to pomoc do nauki języka niemieckiego samodzielny kurs niemieckiego słówka niemieckie na płytach cd w formacie mp3