

## Chemistry - Unit 6 Objectives Describing Substances

1. Classify pure substances and mixtures and molecular and ionic compounds. Describe the structures of pure substances and of mixtures. a. Draw space-filling models for: 1. pure substances with a crystal 2. liquid molecules 3. molecular and ionic structures as pure substances or mixtures	a. pure substances b. c. crystal lattices 1. ionic 2. molecular 3. ionic 4. ionic 5. ionic	10 10 10 10 10
2. Identify physical states of compounds and mixtures. a. List 5 substances that represent 1. mixtures	a. b. c. d.	
3. Compare the chemical and physical properties and chemical behavior of: a. Chlorine with other halogens as physical and chemical behavior b. Chlorine with other halogens as 1. physical and chemical properties	a. chemical properties of gases Boiling point Melting point Density b. water c. density Boiling point Melting point Density Manufacturing	10 10 10 10 10 10 10 10 10 10 10 10
4. Characterize pure substances and mixtures based on their boiling and melting points. Describe the behavior of pure substances and mixtures in their boiling and melting points. a. Compare 5 pure substances b. Explain how the law applies to	a. b.	

# Chemistry Unit 4 Objectives Describing Substances

**Lei Huang**



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