

Chemistry – Unit 6 Notes 1 - Thomson Model of the Atom

J. J. Thomson performed experiments with cathode rays in an attempt to understand electricity – which was still a mystery in the late 1800s. Use the website *A Look Inside the Atom*¹ to find the conclusions that Thomson and other physicists drew regarding the mysterious cathode rays.

Thomson's 1897 Experiments - state the conclusions Thomson drew from each of his famous cathode ray experiments:

1. **First Experiment:** Thomson directed the beam at an electrometer and tried to separate the evidence of charge from the path of the beam. *What connection did Thomson find between charge and the cathode rays? Was the charge positive or negative?*
2. **Second Experiment:** Thomson tried passing the cathode ray through an electric field. *How did cathode ray beam behave when it passed through an electric field? What did he conclude after his second experiment?*
3. **Third Experiment:** Thomson did some careful measurements on how much the path of the cathode ray was bent in a magnetic field and how much energy they carried. From this work Thomson could describe the mass/charge ratio of the cathode ray particles. *What amazing result did Thomson find?*

Thomson's Atomic Model: Thomson presented three hypotheses from his experiments. Only two were accepted by physicists – in fact the third was shown to be wrong! From the first two came a model of the atom known as the *Plum Pudding* model. Complete the atom drawing below to illustrate Thomson's plum pudding model. Explain how this fits with his observations.



Chemistry Unit 6 Modeling Instruction

Liyong Dong



Chemistry Unit 6 Modeling Instruction:

Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Chemistry Unit 6 Modeling Instruction**

In a world inundated with displays and the cacophony of immediate conversation, the profound energy and mental resonance of verbal art often disappear in to obscurity, eclipsed by the continuous onslaught of noise and distractions. Yet, set within the lyrical pages of **Chemistry Unit 6 Modeling Instruction**, a captivating perform of literary splendor that pulses with raw emotions, lies an unique trip waiting to be embarked upon. Written with a virtuoso wordsmith, that exciting opus guides readers on an emotional odyssey, gently exposing the latent possible and profound influence embedded within the complicated web of language. Within the heart-wrenching expanse of the evocative analysis, we will embark upon an introspective exploration of the book is central themes, dissect their fascinating writing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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