

## CHAPTER 21

### TRANSITION METALS AND COORDINATION CHEMISTRY

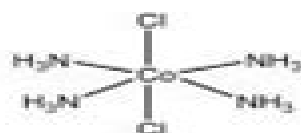
#### Questions

5.  $\text{Fe}_2\text{O}_3(\text{s}) + 6 \text{H}_2\text{C}_2\text{O}_4(\text{aq}) \rightarrow 2 \text{Fe}(\text{C}_2\text{O}_4)_3^{2-}(\text{aq}) + 3 \text{H}_2\text{O}(\text{l}) + 6 \text{H}^+(\text{aq})$ ; the oxalate anion forms a soluble complex ion with iron in rust ( $\text{Fe}_2\text{O}_3$ ), which allows rust stains to be removed.
6. Only the  $\text{Cr}^{2+}$  ion can form four different compounds with  $\text{H}_2\text{O}$  ligands and  $\text{Cl}^-$  ions. The  $\text{Cr}^{2+}$  ion could form only three different compounds, while the  $\text{Cr}^{3+}$  ion could form five different compounds.

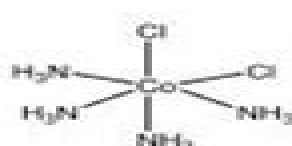
The  $\text{Cl}^-$  ions that form precipitates with  $\text{Ag}^+$  are the counter ions, not the ligands in the complex ion. The four compounds and mol  $\text{AgCl}$  precipitate that would form with 1 mol of compound are:

Compound	Mol $\text{AgCl}(\text{s})$
$[\text{Cr}(\text{H}_2\text{O})_6]\text{Cl}_3$	3 mol
$[\text{Cr}(\text{H}_2\text{O})_5\text{Cl}]\text{Cl}_2$	2 mol
$[\text{Cr}(\text{H}_2\text{O})_4\text{Cl}_2]\text{Cl}$	1 mol
$[\text{Cr}(\text{H}_2\text{O})_3\text{Cl}_3]$	0 mol

7.



trans  
(mirror image is  
superimposable)



cis



The mirror image of the cis  
isomer is also superimposable.

No; both the trans and the cis forms of  $\text{Co}(\text{NH}_3)_4\text{Cl}_2^+$  have mirror images that are superimposable. For the cis form, the mirror image only needs a  $90^\circ$  rotation to produce the original structure. Hence neither the trans nor cis form is optically active.

8. The transition metal ion must form octahedral complex ions; only with the octahedral geometry are two different arrangements of d electrons possible in the split d orbitals. These two arrangements depend on whether a weak field or strong field is present. For four

# Chemquest 21 Transition Metals Answers

**Ajay Kumar Mishra**



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**The Chemistry of the Transition Elements** Alan Earnshaw, Thomas John Harrington, 1973 For freshmen chemistry students The Transition Elements Mary-Lane Kamberg, 2009-08-15 The 37 transitional elements include a huge number of elements that humans use even require on a daily basis The Transitional Elements explores the discovery uses classification and compounds and the impact of these elements on the individual reader Chemistry Of Transition Elements M. Satake & Y. Mido, 2010 Contents The Properties of Transition Elements Titanium Zirconium and Hafnium Group IV A Vanadium Niobium and Tantalum Group V A Chromium Molybdenum and Tungsten Group VI A Manganese Technetium and Rhenium Group VII A Iron Cobalt and Nickel The Platinum Metals Copper Silver and Gold Group IB Analytical and Biological Aspects of Transition Metals Coordination Compounds Lanthanides Actinides Concepts in Transition Metal Chemistry Eleanor Crabb, Elaine Moore, Lesley Smart, 2010 An introduction to transition metals examining the behaviour of the metals and their aqueous ions and complexes complete with The Periodic Table DVD **Reburying the Past** Elizabeth Weiss, Bence Varga, Levente Kis, 2008 Transition metals are the 40 chemical elements 21 to 30 39 to 48 71 to 80 and 103 to 112 The name transition comes from their position in the periodic table of elements In each of the four periods in which they occur these elements represent the successive addition of electrons to the d atomic orbitals of the atoms In this way the transition metals represent the transition between group 2 elements and group 13 elements This book discusses the preparation of a diverse range of transition metal based compounds along with their structural and physical properties their use in chemical synthesis and their role in naturally occurring systems **Transition Metals, Quantitative Kinetics and Applied Organic Chemistry** Brian Chapman, Alan Jarvis, 2004 The revised edition of the highly successful Nelson Advanced Science series for A Level Chemistry Transition Metals Quantitative Kinetics and Applied Organic Chemistry provides full content coverage of Unit 5 of the AS and A2 specifications **Transition Metals - Part 2** Martin J. Mays, 1972 Transition Metals in the Synthesis of Complex Organic Molecules Louis S. Hegedus, 1999 This second edition offers easy access to the field of organotransition metal chemistry The book covers the basics of transition metal chemistry giving a practical introduction to organotransition reaction mechanisms *Practical Chemistry* Mesay Solomon Tesema, Digafie Zeleke, 2024-11-18 This laboratory manual offers a broad introduction to the chemistry of transition elements and more specifically to the chemistry of titanium vanadium chromium molybdenum manganese iron cobalt nickel copper zinc cadmium and mercury The book includes preparation and properties of these transition metals and introduces the chemistry student to the laboratory skills required for accurate and precise chemical analysis **Groups IV, V, and VI Transition Metals and Compounds** T. F. Connolly, 1972-07 responsibility To Betty Edwards and Emily Copenhaver my thanks for what must have seemed endless typing retyping and correcting of these bibliographies over a span of years Availability of Documents U S Government contractor reports usually identified by an alpha numeric report number can be purchased from National

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transition metal nanoparticles synthesised by ion implantation the structural and magnetic characterisation of Cu Picolinate and Cu Quinaldinate molecular systems application of transition metals as active compounds in separation techniques the reactivity of unstable chemicals in the presence of transition metals and the bioinorganic and redox reactions in biological systems of transition metals      *The Heavy Transition Elements* Dr Simon Cotton, S. A. Cotton, Frank Alan Hart, 1975      **The Heavy Transition Elements** S. A. Cotton, Frank Alan Hart, 1975

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