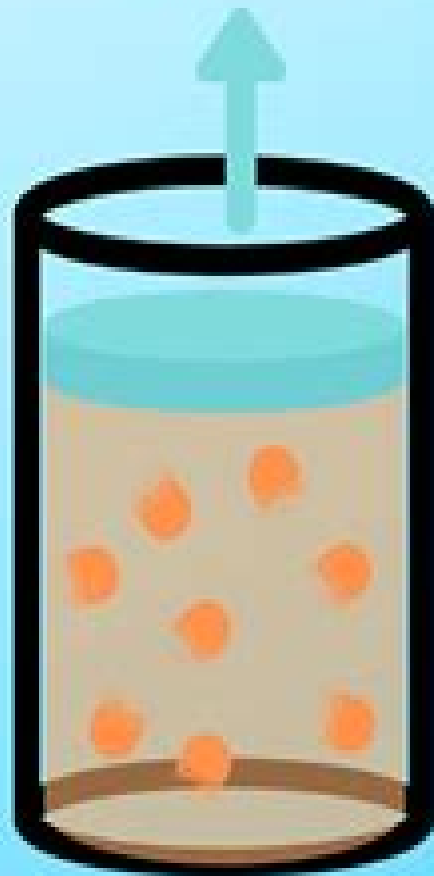


Boyle's Law

The pressure of a gas increases as its volume decreases, assuming constant mass and temperature.

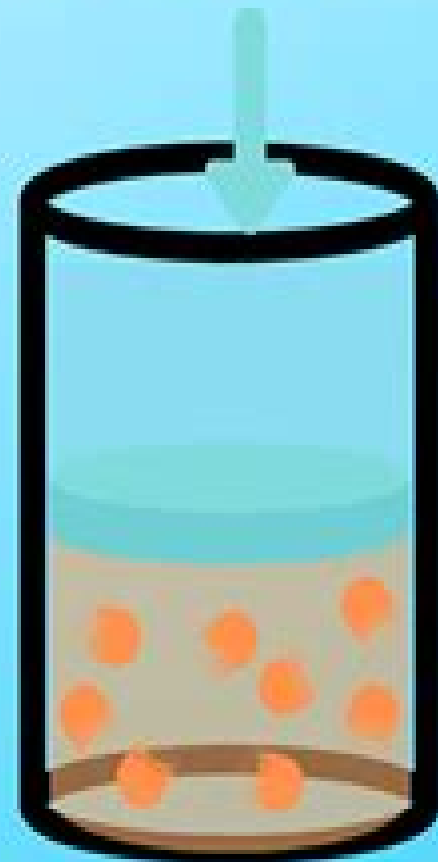


$$P \propto 1/V$$

$$P_1 V_1 = P_2 V_2$$

Pulling up increases volume and decreases pressure.

Pushing down decreases volume and increases pressure.



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RS Peters



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Chemistry and Our Universe , In the first of two lectures on the properties of gases review the basic equations that describe their behavior Learn the history of Boyle s law Gay Lussac s law Charles s law and Avogadro s law Then use these four expressions to derive the celebrated ideal gas law

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