

Charlotte's Tutorial - Isotopes
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 Read it, do it, and show it to me!

Charlotte's Tutorial - Isotopes

1. Record the number of protons, neutrons, and electrons each of the two known isotopes:

	Isotope 1	Isotope 2
Protons	11	11
Neutrons	10	12
Electrons	11	11

2. What does the number of protons determine about an atom?

a. It determines the element. (11 = Sodium)

3. What does the number of protons determine about an atom?

a. It determines the element.

4. What does the number of neutrons determine about an atom?

a. It determines the mass of the atom.

5. Which of the two isotopes will be heavier? Which one?

- a. Isotope 1
- b. Isotope 2
- c. They are the same

6. Chemically identical atoms with different masses are called _____.

Isotopes

7. Label each of the parts of the isotopes equation for sodium described below:

11
23
Na

8. What is the isotope symbol for the other atom of sodium? Fill in the blanks:

11
24
Na

Chemthink Isotopes Answers

Hari Jeevan Arnika



Chemthink Isotopes Answers:

Isotopes in the Atomic Age Hari Jeevan Arnikar, 1989 An account of isotopes separation processes this textbook has been specifically written for student and professional chemical engineers The guide outlines important methods how to improve their yields with discussions of the theory and application of each technique *Isotope Effects in Chemical Reactions* Clair J. Collins, 1971

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