

Molecular Shapes

Questions

1. What steps are involved in predicting molecular shape?
2. Recall VSEPR rules, lone pairs and bonding pairs.
3. Recall the approximate bond angles between bonding pairs.
4. Explain why lone pairs affect the geometry.
5. Drawing the structure of H_2SO_4 is the intermediate step. Drawing the final shape is the final step in the process.

6. Which two molecules listed are shown a deviation from the standard geometry? Between the two molecules:

- a. Which one shows the lone pairs on the central atom? Is it a lone pair or a bond pair?
- b. Which one shows the lone pairs on the central atom? Is it a lone pair or a bond pair?



- a. 180°
 aa. lone pair
 ab. bond pair
 ac. 180°
- b. 180°
 aa. lone pair
 ab. bond pair
 ac. 180°
 ad. 90°



- c. 180°
 aa. lone pair
 ab. bond pair
 ac. 180°
 ad. 90°



- d. Draw the structure of the central atom in the molecule shown. The central atom is the central atom in the molecule.



aa. Carbon dioxide

ab. Carbon dioxide

Chemthink Molecular Shapes Answers

Alison Rodger, Mark Rodger



Chemthink Molecular Shapes Answers:

Molecular Geometry Alison Rodger, Mark Rodger, 1995 Molecular Geometry discusses topics relevant to the arrangement of atoms The book is comprised of seven chapters that tackle several areas of molecular geometry

Shape in

Chemistry Paul G. Mezey, 1993

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