

# CHAPTER 14 STUDY GUIDE

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- |              |   |
|--------------|---|
| <u>false</u> | 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.                               |
| <u>true</u>  | 2. The most abundant substance in a colloid is the dispersion medium.   |
| <u>false</u> | 3. A colloid can be separated by filtration.  |
| <u>true</u>  | 4. A solid emulsion consists of a liquid dispersed in a solid.  |
| <u>true</u>  | 5. Whipped cream is an example of a foam.   |
| <u>false</u> | 6. In an aerosol, the dispersing medium is a liquid.  |
| <u>true</u>  | 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.                |
| <u>true</u>  | 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces. |
| <u>false</u> | 9. Stirring an electrolyte into a colloid stabilizes the colloid.   |
| <u>true</u>  | 10. Colloids demonstrate the Tyndall effect.  |

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		✓	✓
12. Between 1 nm and 1000 nm in diameter	✓	✓	✓
13. More than 1000 nm in diameter	✓	✓	
14. Settle out if undisturbed			
15. Pass through standard filter paper	✓		
16. Lower vapor pressure			✓
17. Scatter light	✓		

# Chemistry Study Guide Mixtures And Solutions

**Floyd G. Robinson**



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