

Describing Substances Review

1. Classify each of the pictures below by placing the correct label in the blanks below.

A = Element

B = Compound

C = Mixture of Elements

D = Mixture of Compounds

E = Mixture of Elements and Compounds



a. B



b. C



c. E



d. D



e. A



f. B



g. A



h. E



i. C

2. A mixture (is/**is not**) a chemical combining of substances.
3. In a compound the (atoms/^{will be on your quiz}molecules) are (**chemically**/physically) combined so that the elements that make up the compound (retain/**lose**) their identities and (**do**/do not) take on a new set of properties.
4. True or **False**: A mixture is always made up of a combination of elements.
5. In a mixture, the substances (lose/**retain**) their identities.
6. In a mixture the substances involved (**can**/cannot) be separated by a simple physical process.
7. In a compound the elements involved (can/**cannot**) be separated by a simple physical process because the elements are (physically combined/**chemically bonded**).
8. True or **False**: An element can be broken down into a simpler substance.

Chemistry Unit 4 Describing Substances 3 Answers

C. J. W. Brooks, A. F. Fell, R. W. Frei



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College Biology I James Hall Zimmerman, Sophie E. Merritt, 1963

Study Guide to Accompany Basics for

Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students Chemistry

Henry Dorin, 1987

The Fluid Earth Teacher's Guide ,

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Approach Burdge, 2016-04-16 Ebook Introductory Chemistry An Atoms First Approach

Holt Chemistry Ralph Thomas

Myers, 2004

Journal of Education , 1916

Reviewing Chemistry Peter E. Demmin, 1989

TRAC: Trends in

Analytical Chemistry C. J. W. Brooks, A. F. Fell, R. W. Frei, 2013-09-03 Trends in Analytical Chemistry Volume 5 focuses on the advancements of processes technologies automation and applications of analytical chemistry The selection first offers information on graphics programming for the IBM PC using FORTRAN PASCAL and C including graphics hardware system software assembly language routines and high level interface The text then elaborates on the place of affinity chromatography in the production and purification of biomolecules from cultured cells and zone electrophoresis in open tubular capillaries Discussions focus on column and instrument design applications affinity chromatography in protein production from cells and economic aspects of production and purification of proteins from cell cultures The manuscript takes a look at polarographic and voltammetric techniques and their application to the determination of vitamins and coenzymes and activation analysis with charged particles Topics include accelerators principle of charged particle activation analysis and applications The text then examines the development of microbiological and immunological assays for

antibiotics and the use of computer system for a small analytical research laboratory The book is a dependable reference for readers interested in the trends in analytical chemistry **Chemistry International** ,1994 **Engineering Journal** ,1929

The Medical World ,1900 **Smith's Inorganic Chemistry** Alexander Smith,James Kendall,1926 Education Outlook ,1916 **Educational Times** ,1916 **Educational Times and Journal of the College of Preceptors** ,1909

The Educational Times, and Journal of the College of Preceptors ,1916 **Frank Schaffer's Chemistry for Everyday** David Thurlo,1999 Students learn about important subjects by relating them to events and things that occur in their everyday lives A wealth of interesting activities provide a detailed look into each subject Easy to use activities can be completed individually at school or at home though a few hands on experiments require group work and data sharing A great supplement to any existing curriculum Includes topics such as the scientific method applied to chemistry determining specific gravity balancing chemical equations and exploring the periodic table of elements Science Examinations ... Reports, Etc ,1906 **Engineering and Mining Journal** ,1901

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