





## Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being their predictable characteristic chemical behaviour. Atoms of one element react in a predictable way with atoms of another element. However, when the elements react, sometimes several particles are combined to form compounds. They do not combine randomly as solids. We will now examine the patterns that exist for the ratios in which these elements combine to form the compounds that emerge at the end of the reaction.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 1 oxygen
	$\frac{12}{16}$	$\frac{12}{16}$	$\frac{12}{16}$	$\frac{1}{16}$
particle diagram				
formula				

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen
	$\frac{12}{16}$	$\frac{12}{16}$	$\frac{12}{16}$
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formula			

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 16 oxygen	1 hydrogen 16 oxygen
	$\frac{12}{16}$	$\frac{12}{16}$	$\frac{12}{16}$	$\frac{1}{16}$
particle diagram				
formula				

# Chemistry Unit 6 3 Answers

**Liyong Dong**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Block Elements 9 General Organic Chemistry 10 Hydrocarbons Halogen Derivatives 11 Colligative Properties Of Solution 12 Chemical Kinetic Nuclear Chemistry 13 Solid State Surface Chemistry Colloids 14 Electrochemistry 15 The P Block Elements 16 The Transition Co Ordination Compounds 17 Analytical Chemistry 18 Compound Contains Oxygen 19 Compound Contains Nitrogen Practical Organic Chemistry 20 Carbohydrates Amino Acid Misc Match The Following 21 Model Test Paper

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