

Chemistry Worksheets Class 11 on Chapter 5 States of Matter- Set 1

Q-1: At 600 K, the vapour pressure of water in a closed container is 0.5 atm. If the container's volume is doubled, its vapour pressure at 600 K will be

- a) 0.2 atm
- b) 0.8 atm
- c) 0.6 atm
- d) 0.5 atm

Q-2: In the equation $PV = RT$, the value of R will depend on

- a) The nature of the gas
- b) The pressure of the gas
- c) The temperature of the gas
- d) Units of measurement

Q-3: Air at sea level is dense. This is the practical application of

- (a) Charles's law
- (b) Boyle's law
- (c) Dalton's law
- (d) Avogadro's law

Q-4: Triple point of water is

- a) 373.16 K
- b) 273.16 K
- c) 193.14 K
- d) 203.14 K

Q-5: The slope of the plot between PV and P at constant temperature is

- a) 1
- b) Zero
- c) $\frac{1}{2}$
- d) None of the above

Q-6: While pumping air into a cycle tyre, a boy noticed that volume, as well as pressure, was increasing. Does Boyle's law fail here?

Chemistry Worksheet Chapter 5 Concept Review Answer Sheet

Rex M. Heyworth



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1. Understanding the eBook Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - The Rise of Digital Reading Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Personalized Recommendations
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet User Reviews and Ratings
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet and Bestseller Lists
5. Accessing Chemistry Worksheet Chapter 5 Concept Review Answer Sheet Free and Paid eBooks
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet Public Domain eBooks
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet eBook Subscription Services
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet Budget-Friendly Options

6. Navigating Chemistry Worksheet Chapter 5 Concept Review Answer Sheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet Compatibility with Devices
 - Chemistry Worksheet Chapter 5 Concept Review Answer Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Highlighting and Note-Taking Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Interactive Elements Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
8. Staying Engaged with Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
9. Balancing eBooks and Physical Books Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Setting Reading Goals Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Fact-Checking eBook Content of Chemistry Worksheet Chapter 5 Concept Review Answer Sheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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