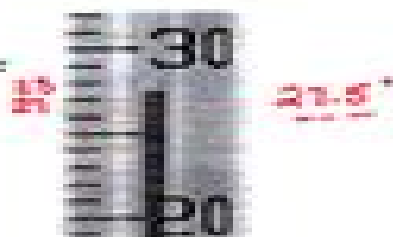


1 • Matter and Measurement

1. How many significant digits are present in the temperature read from the thermometer illustrated to the right?



- a) 1 b) 2 **c) 3** d) 4

2. The dimensions of a rectangular solid are 8.00 cm long, 4.00 cm wide, and 2.00 cm high. If the density of the solid is 10.0 g/cm³, what is its mass?

- a) 1064 grams d) 320 grams
b) 10.0 grams **c) 640 grams** (3 sig figs)
c) 64.0 grams

$$V = l \times w \times h = 64.0 \text{ cm}^3$$

3. A metal sample weighing 30.9232 grams was added to a graduated cylinder containing 23.26 mL of water. The volume of water plus the sample was 24.85 mL. Which setup will result in the density of this metal?

- a) $30.9232 \times (24.85 - 23.26)$

b) $\frac{30.9232}{24.85 - 23.26}$

c) $\frac{24.85 - 23.26}{30.9232}$

d) $30.9232 \times \frac{24.85}{23.26}$

e) $\frac{30.9232}{24.85 + 23.26}$

$$V = 24.85 - 23.26 = 1.59$$

$$D = \frac{m}{V} = \frac{30.9232}{1.59}$$

PRACTICE TEST

4. The number of significant digits in 0.00001 is

- a) 1 d) 4
b) 2 **c) 5**
c) 3

5. A box measures 3.50 cm x 2.515 cm. The product of these numbers = 10.2025 cm². What is the proper way to report the area of the box?

- a) 10.20 cm² c) 10 cm²
b) 10.2 cm² d) 10 cm²

6. The result of $2.358 \times (4.0 + 6.311)$ is

- a) 24 c) 24.21
b) 24.2 d) 24.205

A student does a calculation using her calculator and the number 290.27163 is shown on the display. If there are actually three significant figures, how should she show the final answer?

- a) 290 b) 2.90×10^2
c) 2.90×10^3
d) 290.27 e) 290.27163

7. The term that refers to the reproducibility of a laboratory measurement is

- a) precision** c) accuracy
b) repeatability d) exactness

Which measurement below is NOT written with three significant digits?

- a) 2.00 cm** b) 0.003 L
c) 550. grams d) 12.7 mm

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LL Leslie



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