

AP Chemistry Cumulative Test Review

Mole Fraction - Answer The number of moles of a particular substance compared to the total number of moles. $X_a = \text{mole } a / \text{mole } a + \text{mole } b$

Percent Error - Answer -Accepted value-experimental value/accepted value
-Multiply found # by 100

Isotope - Answer -Elements that have different numbers of neutrons
-Atomic Mass-Protons = Neutrons

Molecular Formula - Answer A chemical formula that indicates the actual number of atoms of each element in one molecule of a substance.

Empirical Formula - Answer A chemical formula that shows the kinds of atoms and their relative numbers in a substance in the smallest possible whole-number ratios.

Hydrate - Answer Ionic compound that naturally contains H₂O molecules incorporated into its crystal structure.

Anhydrate - Answer Ionic compound that naturally contains H₂O molecules incorporated into its crystal structure, but the water has been removed by some chemical or physical means.

Matter tree - Answer A map representing all atoms and molecules

Impurity - Answer A substance is mixed with what you believe to be pure.

s orbital - Answer first electron orbital and is spherical (round)

p orbital - Answer second electron orbital and is dumbbell shaped

d orbital - Answer third electron orbital and balloon shaped

f orbital - Answer fourth electron orbital and irregular in shape

How many electrons can the s orbital hold? - Answer 2

Chemistry Test Cumulative Review Answers

Rachel Sandford



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