

## Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being their predictable characteristic chemical behaviour. Atoms of one element cannot be transformed into atoms of another element, which demonstrates that each element's particles are made up of atoms with only one type of nucleus. However, when the elements are split, subatomic particles are revealed to form compounds along with atoms of other elements, as we will see. We will now examine the patterns that reveal the structure of atoms, which shows almost a complete lack of change in the structure of the basic building blocks.

1. Write the formulae and draw the particle diagrams for each compound. The ratio of atoms in each compound is given.

Atomic structure formulae	12 hydrogen 16 oxygen $\text{H}_2\text{O}$	12 hydrogen 16 oxygen $\text{H}_2\text{O}_2$	12 hydrogen 1 oxygen $\text{H}_2\text{O}_2$	1 hydrogen 1 oxygen $\text{H}_2\text{O}$
particle diagrams				

Atomic structure formulae	12 hydrogen 16 oxygen $\text{H}_2\text{O}$	12 hydrogen 16 oxygen $\text{H}_2\text{O}_2$	12 hydrogen 1 oxygen $\text{H}_2\text{O}_2$
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# Chemistry Unit 6 Lesson 1 Answers Key

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