		Marrie-
	mentative: Unit 10 Authorison Seat Revision	19 Date
90	In a solution, the substance being dissolved is the	
100		355
	The most commonly used solvent is	
9.	Ethylene glycal (antifreede) and water are completely n water in a beaker, what would you expect to see?	nisoble. If you mix 30 ml. of ethylene glycol with 10 ml. of
H	A solution of socium obloride and water is what type of	ineque?
6	The must precise description of water and oil would be to say that these two substances are	
4	Give several examples of an emulsifying agent?	
S	What are the properties of an emulsiying agent? On a	ther words, what do they do? For what are they used?)
60	A silver colored deersii filling is a(x)	and an lodine setution for subs is a(n)
٥.	Give two examples of alloys:	
100	. What are the four factors that affect the rate of dissolu	ation?
in.	Some drugs taken by athletes can show up on drug to These drugs are probably	
u e	e your solubility curve to answer the next the ques	done.
12	Batwase 40°C and 60°C, which compound shows the	largest increase in solubility?
12.	. The solubility of which compound is UEAST affected by temperature?	
14.	The adubility of KNO, increases by approximately	when the temperature rises from 16°C to 60°C.
18.	For 140 g of Ki (in 100 g H <sub>2</sub> O) at 10°C, the estution w	ould be described as being
116.	Which of the following solutions is smallpressed?  80 g of KCII (in 100 g of water) at 86°C  40 g of NaCII (in 100 g of water) at 86°C  146 g of NaKCII (in 100 g of water) at 46°C  45 g of NHLCII (in 100 g of water) at 45°C	
12	A actualization of HCI is 0.300 M. Determine the mass of	HOI reseded to make 150 mL of solution.

16. A 2.35 M solution of Na<sub>2</sub>OO<sub>2</sub> contains 67 g of solute. Find the solute of the solution.

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**Stephen P. Gordon** 

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