

**Naming and Writing Formulas for
Covalent (Molecular) Compounds**

Name: _____

Part A: Name the covalent compounds

1. CO _____
2. CO₂ _____
3. N₂O₃ _____
4. NP _____
5. SCl₂ _____
6. P₂O₅ _____
7. NBr₃ _____
8. Cl₄ _____
9. CCl₄ _____
10. PF₅ _____
11. SeF₂ _____
12. TeBr₂ _____
13. P₂S₅ _____
14. C₃N₄ _____
15. CH₄ _____
16. PH₃ _____

17. Part B: Write the chemical formula

1. carbon tetrafluoride _____
2. silicon dioxide _____
3. dinitrogen trisulfide _____
4. phosphorus mononitride _____
5. carbon disulfide _____
6. dinitrogen trichloride _____
7. silicon tetrabromide _____
8. carbon dioxide _____
9. nitrogen trifluoride _____
10. boron trisulfide _____
11. disulfur trioxide _____
12. selenium tetrafluoride _____
13. diphosphorus pentasulfide _____
14. sulfur dibromide _____
15. carbon tetrachloride _____
16. dinitrogen tetroxide _____

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