

ChemQuest 8: Answer Key

Structure of the Atom

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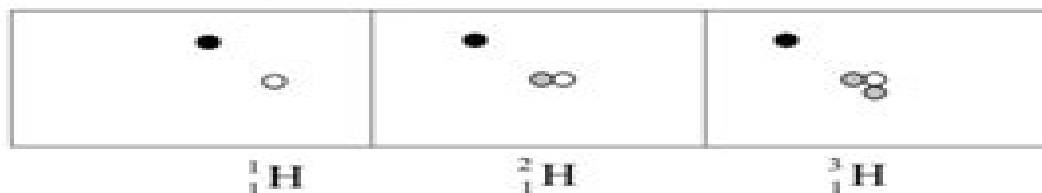
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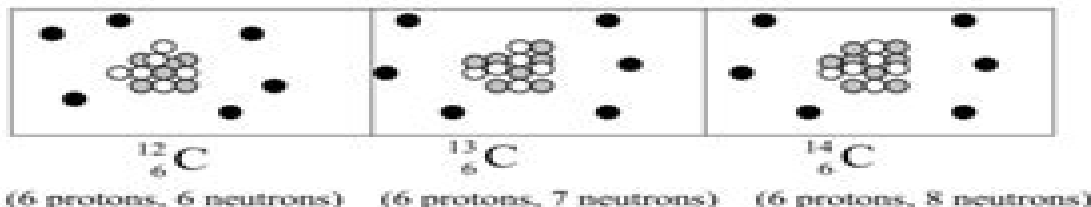
Note the following symbols: (they are not to scale)

- = proton (positive charge)
- = electron (negative charge)
- ⊙ = neutron (no charge)

The following three diagrams are hydrogen atoms:



The following three diagrams are carbon atoms:



Notice the type of notation used for atoms:



X = chemical symbol of the element

Z = "atomic number"

A = "mass number"

${}^{12}_6\text{C}$, ${}^{13}_6\text{C}$, and ${}^{14}_6\text{C}$ are notations that represent isotopes of carbon.

${}^1_1\text{H}$, ${}^2_1\text{H}$ and ${}^3_1\text{H}$ are notations that represent isotopes of hydrogen.

The part of the atom where the protons and neutrons are is called the nucleus.

Chemquest Answer Key

Rachel Turoscy



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Comprehensive Medicinal Chemistry Corwin Hansch, Peter George Sammes, John Bodenhan Taylor, 1990 Drug design is a multi disciplinary activity involving chemists biologists biochemists pharmacologists and many others the chemist's role is central in inventing new compounds which exert a beneficial effect However once a lead for a new active drug has been established its effective delivery has to be demonstrated and extensive toxicological studies undertaken to demonstrate its safety before clinical trials can commence The metabolic fate of the drug has to be revealed and detailed distribution studies carried out in order to satisfy the regulatory authorities before the new compound can be marketed Comprehensive Medicinal Chemistry describes all these aspects of the design of a drug whilst centering on the chemical mechanism whereby such agents act Volume 4 covers quantitative drug design

The Directory of Management Consultants, 2003 Kennedy Information (Firm), 2002

The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume One) (First Edition) Rachel Turoscy, 2018-08-09 The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to problem solving methodology and unit conversions In subsequent chapters the text covers important topics like ionic and covalent bonding chemical formula calculations solubility and reactions in aqueous solution gases the first law of Thermodynamics Quantum theory and electron configuration It also covers periodic trends the Lewis Dot Structures and bonding theories Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry

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