

Chemthink: CHEMICAL REACTIONS

- a) Go to www.chemthink.com. Log in using your user name and password. (If you forgot, you need to WAIT QUIETLY until your teacher is done introducing the lesson.)
- b) Scroll to **CHEMICAL REACTIONS** then the tutorial under the Chemical Reactions menu.
- c) Complete the **tutorial**. As you complete the tutorial, write down the answers to the questions below.
- d) Once you have successfully completed the tutorial and this worksheet, complete the **Question Set**.

CHEMICAL REACTIONS TUTORIAL QUESTIONS:

- Starting materials in a chemical reaction are called _____.
- The ending materials in a chemical reaction are called _____.
- The arrow indicates a _____ has taken place.
- All reactions have one thing in common: there is a _____ of chemical bonds.
- Chemical reactions always involve _____, old bonds, _____, new bonds, or both.
- In all reactions we still have all of the _____ at the end that we had at the start.
- In every reaction there can never be any _____ atoms or _____.
- Chemical reactions only _____ in the atoms that are already there.
- Let's represent a reaction on paper. For example, hydrogen gas (H_2) reacts with oxygen gas (O_2) to form water (H_2O):



If we use only the atoms shown, we'd have _____ atoms of H and _____ atoms of O as reactants. This would make _____ molecule of H_2O , but we'd have _____ atom of O leftover. However, this reaction only makes H_2O .

Remember: reactions are not limited to 1 molecule each of reactants. We can use as many as we need to balance the chemical equation.

A balanced chemical reaction shows:

- a) What atoms are present before (in the reactants) and after (in the products)
 - b) **How many** of each reactant and product is present before and after.
10. So to make H_2O from oxygen gas and hydrogen gas, the balanced equation would be:



Which is the same as:

# of atoms in Reactants	Element	# of atoms in Products
	H	
	O	

- This idea is called the _____.
- There must be the same _____ and the same number of _____ before the reaction (in the reactants) and after the reaction (in the products).
- What is the balanced equation for this reaction? _____ Cu + _____ $O_2 \rightarrow$ _____ CuO

Chemthink Chemical Reactions Questions

Denise Walker



Chemthink Chemical Reactions Questions:

Druggists' Circular and Chemical Gazette, 1869 Chemical Reactions Science Learning Guide NewPath Learning, 2014-03-01 The Chemical Reactions Student Learning Guide includes self directed readings easy to follow illustrated explanations guiding questions inquiry based activities a lab investigation key vocabulary review and assessment review questions along with a post test It covers the following standards aligned concepts Changes of Matter Chemical Reactions Formulas Balancing Equations Types of Chemical Reactions 1 Types of Chemical Reactions 2 Energy in Chemical Reactions Evidence of Chemical Reactions and Chemical Reaction Rates Catalysts Aligned to Next Generation Science Standards NGSS and other state standards **Chemical Reactions and Their Equations** Ingo Waldemar Dagobert Hackh, 1928 **Chemical Reactions** Denise Walker, 2007 This title introduces the reader to the huge variety of chemical reactions that shape our world Find out all about explosions learn about how to start reactions and understand how chemical equations work **Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice)** Heather Hattori, Richard H. Langley, 2014-04-14 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time **CHEMICAL REACTIONS** NARAYAN CHANGDER, 2024-04-08 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquiziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment

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