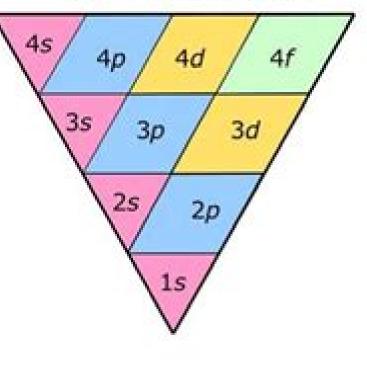
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dots shows that the electron spends more time closer to n = 1 Sublevel designation



The first four principal energy levels in the hydrogen atom. Each level is assigned a principal quantum number n.

The types of orbitals on each of the first four principal energy levels.



An orbital for a hydrogen atom. The intensity of the

the nucleus.

Chemthink Answers Atomic Structure

Jack Challoner

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