

Exercises (100)

Controlling Chemical Reactions

If the statement is true, write "true". If the statement does change the statement word or words to make the statement true.

1. Le Chatelier's Principle states that the equilibrium of a reaction shifts in the direction that counteracts the change in the reaction conditions.
2. True The concentration of a reactant in a closed system is the equilibrium of the reaction.
3. Le Chatelier's Principle states that the effect of a change in a reaction condition is to counteract the change.
4. True Le Chatelier's Principle states that a reaction shifts in the direction that counteracts the change in the reaction conditions.
5. True Le Chatelier's Principle states that a reaction shifts in the direction that counteracts the change in the reaction conditions.

Put in the blank a complete word or statement.

1. The reaction of a gas with a solid catalyst gives a higher rate of reaction.
2. In an equilibrium reaction, the change in the concentration of a reactant is the same as the change in the concentration of a product.
3. Increasing the concentration of a reactant in a reaction shifts the equilibrium to the right.
4. The amount of a substance that is produced in a reaction is the same as the amount of a reactant that is consumed.
5. The amount of a substance that is produced in a reaction is the same as the amount of a reactant that is consumed.

Controlling Chemical Reactions Lesson Quiz Answers

Clifford Lane



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Chapter Resource 39 Digestive/Excretory Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004
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Chemical & Process Engineering, 1964 *The Chemical News*, 1919 *Chemical News and Journal of Industrial Science*, 1918

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