

Chemistry Worksheets Class 11 on Chapter 5 States of Matter- Set 1

Q-1: At 600 K, the vapour pressure of water in a closed container is 0.5 atm. If the container's volume is doubled, its vapour pressure at 600 K will be

- a) 0.2 atm
- b) 0.8 atm
- c) 0.6 atm
- d) 0.5 atm

Q-2: In the equation PV = RT, the value of R will depend on

- a) The nature of the gas
- b) The pressure of the gas
- c) The temperature of the gas
- d) Units of measurement

Q-3: Air at sea level is dense. This is the practical application of

- (a) Charle's law
- (b) Boyle's law
- (c) Dalton's law
- (d) Avogadro's law

Q-4: Triple point of water is

- a) 373.16 K
- b) 273.16 K
- c) 193.14 K
- d) 203.14 K

Q-5: The slope of the plot between PV and P at constant temperature is

- a) 1
- b) Zero
- c) ½
- d) None of the above

Q-6: While pumping air into a cycle tyre, a boy noticed that volume, as well as pressure, was increasing. Does Boyle's law fail here?

<u>Chemistry Worksheet Chapter 5 Concept Review Answer</u> <u>Sheet</u>

Clemens Wendtner

Chemistry Worksheet Chapter 5 Concept Review Answer Sheet:

tools and activities to support differentiated instruction

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