Name Key

Class

Date

# **Nuclear Decay**

### Types of Radiation

Complete the table below to compare the properties of alpha, beta, positron, and gamma radiation.

Name	Alpha	Beta	Positron	Gamma
Greek letter	α	β-	β+	Y
Symbol	*He	e 1e	,*e	N/A
Composition	2p°, 2n°	electron	anti-electron	EM wave
Charge	2+	1-	1+	0
Stopped by	paper, skin, clothing	wood, glass	wood, glass	thick concrete or lead

### **Nuclear Decay Reactions**

In problems 2 - 5, fill in the missing particle for each nuclear decay reaction. Then, match each equation to the appropriate type of nuclear decay.

$$B_2 = {}^{14}C \rightarrow {}^{14}N + {}^{6}e$$

$$\underline{D}$$
 3.  $^{32}_{17}CI \rightarrow ^{0}_{+1}e + ^{32}_{16}S$ 

$$A$$
 4.  ${}^{218}_{84}Po \rightarrow {}^{4}_{2}He + {}^{214}_{82}Pb$ 

$$C_5$$
.  $^{142}Pm + ^{0}_{-1}e \rightarrow ^{142}Nd$ 

A. alpha emission

B. beta emission

C. electron capture

D. positron emission

In problems 6 and 7, write a balanced equation for each nuclear decay reaction.

- Decay of radium-223 by alpha (α) emission.
- Decay of iodine-126 by positron (β+) emission.

$$^{223}_{88}Ra \rightarrow ^{4}_{2}He + ^{219}_{86}Rn$$

$$^{126}_{53}I \rightarrow ^{0}_{-1}e + ^{126}_{52}Te$$

8. How are the atomic number and mass number of a nuclide affected by positron emission?

Atomic number decreases by one. Mass number stays the same.

- How are the atomic number and mass number of a nuclide affected by the emission of gamma radiation?
   Both atomic number and mass number stay the same since gamma radiation has no mass or charge.
- 10. Which has a slightly greater mass in a nuclear reaction, the reactants or the products? Why? The reactants have a slightly greater mass. In a nuclear reaction, a small amount of mass is converted to energy according to the equation E = mc². The difference in mass is referred to as the mass defect.

# **Chemistry Worksheet Questions 1 On Radioactivity Answers**

**R Pring** 

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# Chemistry Worksheet Questions 1 On Radioactivity Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "Chemistry Worksheet Questions 1 On Radioactivity Answers," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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# **Table of Contents Chemistry Worksheet Questions 1 On Radioactivity Answers**

- 1. Understanding the eBook Chemistry Worksheet Questions 1 On Radioactivity Answers
  - The Rise of Digital Reading Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Worksheet Questions 1 On Radioactivity Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Personalized Recommendations
  - Chemistry Worksheet Questions 1 On Radioactivity Answers User Reviews and Ratings

- Chemistry Worksheet Questions 1 On Radioactivity Answers and Bestseller Lists
- 5. Accessing Chemistry Worksheet Questions 1 On Radioactivity Answers Free and Paid eBooks
  - Chemistry Worksheet Questions 1 On Radioactivity Answers Public Domain eBooks
  - Chemistry Worksheet Questions 1 On Radioactivity Answers eBook Subscription Services
  - Chemistry Worksheet Questions 1 On Radioactivity Answers Budget-Friendly Options
- 6. Navigating Chemistry Worksheet Questions 1 On Radioactivity Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Chemistry Worksheet Questions 1 On Radioactivity Answers Compatibility with Devices
  - Chemistry Worksheet Questions 1 On Radioactivity Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Highlighting and Note-Taking Chemistry Worksheet Questions 1 On Radioactivity Answers
  - o Interactive Elements Chemistry Worksheet Questions 1 On Radioactivity Answers
- 8. Staying Engaged with Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Worksheet Questions 1 On Radioactivity Answers
- 9. Balancing eBooks and Physical Books Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Worksheet Questions 1 On Radioactivity Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Setting Reading Goals Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Fact-Checking eBook Content of Chemistry Worksheet Questions 1 On Radioactivity Answers
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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