

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: \uparrow temp = \uparrow speed

type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

\downarrow
b/c molecules moving faster

$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

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Table of Contents Chemthink Gases Questions Answers

1. Understanding the eBook Chemthink Gases Questions Answers
 - The Rise of Digital Reading Chemthink Gases Questions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemthink Gases Questions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemthink Gases Questions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemthink Gases Questions Answers
 - Personalized Recommendations
 - Chemthink Gases Questions Answers User Reviews and Ratings
 - Chemthink Gases Questions Answers and Bestseller Lists
5. Accessing Chemthink Gases Questions Answers Free and Paid eBooks

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 - Chemthink Gases Questions Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Chemthink Gases Questions Answers
 - Interactive Elements Chemthink Gases Questions Answers
- 8. Staying Engaged with Chemthink Gases Questions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemthink Gases Questions Answers
- 9. Balancing eBooks and Physical Books Chemthink Gases Questions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemthink Gases Questions Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemthink Gases Questions Answers
 - Setting Reading Goals Chemthink Gases Questions Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemthink Gases Questions Answers
 - Fact-Checking eBook Content of Chemthink Gases Questions Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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