

CHAPTER **11** STUDY GUIDE

Stoichiometry

Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- _____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- _____ 2. Stoichiometry is based on the law of conservation of mass.
- _____ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- _____ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- _____ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.

methanol + oxygen \rightarrow carbon dioxide + water



Substance	Molar Mass (g/mol)	Number of Molecules	Number of Moles (mol)	Mass (g)
6. Methanol	32.05			
7. Oxygen gas	32.00			
8. Carbon dioxide	44.01			
9. Water	18.02			

10. What are the reactants? _____
11. What are the products? _____
12. What is the total mass of the reactants? _____
13. What is the total mass of the products? _____
14. How do the total masses of the reactants and products compare? _____

Chapter 11 Study Guide For Content Mastery Section 11 1 Measuring Matter Answers

James Augustus Henry Murray



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Educational Times ,1886 **Walford's Concise Guide to Reference Material** Albert John Walford,1992 This is a shortened version of the three volume Walford's Guide to Reference Material 5th edition Volume 1 Science and Technology 1989 Volume 2 Social and historical sciences philosophy and religion 1990 and Volume 3 Generalia language and literature the arts 1991 There are more than 3 000 entries forming an updated compilation of what are considered to be the basic items in the main volumes plus some more recent material up to April 1992 *Physics Briefs* ,1979 **The Oxford English Dictionary** ,1989 In addition to current definitions provides an historical treatment to words and idioms included

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he also requested arjuna and krishna to take benedictions as he was very much pleased by their valour arjuna asked indra for many divine weapons and mantras to control them indra granted them all he asked krishna to take some benediction krishna asked the following beautiful benediction and it is worth meditating for all aspiring devotees

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