ANATOMY & PHYSIOLOGY

CHAPTER 2 STUDY GUIDE

SHOW REPORT OF

Chemistry of Life

Levels of Chemical Organization.

- Atoms: The basic unit of matter, containing a nucleus made up of protons (positively charges) and neutrons (uncharges).
 - Atomic Number: Number of protons in the nucleus; defines the element.
 - Attentic Mass: Total number of protons and neutrons.
 - Exectrons: Negatively charged particles that orbit the nucleus in energy levels forbitals), each consider of holding up to eight electrons.

Elements, Molecules, and Compounds

- Element: A pure substance made up of only one kind of atoms.
- · Molecula: A group of atoms bonded tagether.
- Compound: A substance with exploration that contain more than one kind of store.

Chamical Bonding

- Purpose: Bonds form to stabilize atoms by filling their outer energy levels.
- Terres of Bonds:
 - tank Bands: Formed by the attraction of appositely changed lons.
 - Pasitive los: Atom that loses electrons (e.g., No., Ca²).
 - Megative foe: After that gains electrons (e.g., CF).
 - Covalent Bonds: Formed by the sharing of electrons between atoms; generally stable and do not easily dissociate in water.
 - Hydrogen Bands: Weak attractions that hold molecules together in folded shapes or groups; do not create new molecules.

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AW Rasmussen



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