

12

STOICHIOMETRY

Practice Problems

In your notebook, solve the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups (top and bottom), and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate, $\text{KClO}_3(\text{s})$. Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from 3.8×10^{23} molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(II) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 75.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. Write a balanced equation first.

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of dinitrogen monoxide. Assume all gases are at the same conditions of temperature and pressure.



Chapter 12 Stoichiometry Practice Problems Answer Key

Melissa Ann Flynn



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Addison-Wesley Chemistry, 1990

Soil and Groundwater Remediation Chunlong Zhang, 2019-10-30 An introduction to the principles and practices of soil and groundwater remediation Soil and Groundwater Remediation offers a comprehensive and up to date review of the principles practices and concepts of sustainability of soil and groundwater remediation The book starts with an overview of the importance of groundwater resource quality contaminant sources types and the scope of soil and groundwater remediation It then provides the essential components of soil and groundwater remediation with easy to understand design equations calculations and the practical applications The book contains information on remediation basics such as subsurface chemical behaviors soil and groundwater hydrology and characterization regulations cost analysis and risk assessment The author explores various conventional and innovative remediation technologies including pump and treat soil vapor extraction bioremediation incineration thermally enhanced techniques soil washing flushing and permeable reactive barriers The book also examines the modeling of groundwater flow and contaminant transport in saturated and unsaturated zones This important book Presents the current challenges of remediation practices Includes up to date information about the low cost risk based sustainable remediation practices as well as institutional control and management Offers a balanced mix of the principles practices and sustainable concepts in soil and groundwater remediation Contains learning objectives discussions of key theories and example problems Provides illustrative case studies and recent research when remediation techniques are introduced Written for undergraduate seniors and graduate students in natural resource earth science environmental science engineering and environmental management Soil and Groundwater Remediation is an authoritative guide to the principles and components of soil and groundwater remediation that is filled with worked and practice problems

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