

Chapter 3**ENRICHMENT**

Use with Section 1

• Introduction to the Periodic Table

Mendeleev's Periodic Table

The modern periodic table is based on the work of Dmitri Mendeleev, a nineteenth-century Russian chemist and professor. In the 1860s, while working on writing a chemistry textbook, he came up with a way to organize the elements by their atomic weights. He arranged the elements into groups with the same properties. What you see here is the table he had developed by 1871.

Answer the following questions, using complete sentences.

I	II	III	IV	V	VI	VII	VIII
H							
Li	Be	B	C	N	O	F	
Na	Mg	Al	Si	P	S	Cl	
K	Ca	—	Ti	V	Cr	Mn	Fe, Co, Ni
Cu	Zn	—	—	As	Se	Br	Ru, Rh, Pd
Ag	Cd	In	Sn	Sb	Te	I	
Cs	Ba						

- How is this table different from the modern periodic table? _____

- Why do you think some elements are missing from the table?

- Mendeleev left a few blanks in the table (marked above with a dash). He knew that any spot on the table could only be filled by an element with a particular atomic weight and properties. He inferred that some elements hadn't yet been discovered, since no known elements at the time seemed to fit these blank spots. He expected that one of the spots would someday be filled by an element with properties similar to silicon (Si). Which spot do you think he had chosen for this element? Explain.

Dmitri Mendeleev Worksheet Answers

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



Dmitri Mendeleev Worksheet Answers:

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket

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Holt Chemistry Holt Rinehart & Winston,2003-01-24 *Merrill Chemistry* Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998

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