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Chemistry Chapter 12: Stoichiometry Study Guide

Section 12.1 What is stoichiometry?

1. A flame burning is a _____ between the fire and the air in the room
 - a. Explain this reaction

2. In many cases, reactions will keep going until one of the _____ is used up.

3. The study of quantitative relationships between amounts of reactants used and products formed by a chemical reaction is called _____.
 - a. What does this mean?

4. **Stoichiometry** is based around the concept of the _____.
 - a. What does it state?

 - b. We oftentimes end up with different things than we started with but the amount of _____ present at the end of the reaction is the same as at the beginning

5. Stoichiometry is understanding the relationship between the _____ and the _____

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