% Concentration

Example Problem 1 (Solid in a Liquid)

What is the concentration in %w/v of a solution containing 39.2 g of potassium nitrate in 177 mL of solution?

% (w/v) =
$$\frac{mass\ solute}{volume\ solution} \times 100$$
 $\frac{39.2g}{177mL} \times 100 = 22.1$ % w/v

Example Problem 2 (Liquid in a Liquid)

What is the concentration in %v/v of a solution containing 3.2 L of ethanol in 6.5 L of solution?

% (v/v) =
$$\frac{volume\ solute}{volume\ solution} \times 100$$
 $\frac{3.2L}{6.5L} \times 100$ = 49 % v/v

Chapter 15 Section 152 Solution Concentration Answers

Sir Henry Wilmot Seton

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